

'Control of anchoring of discotic liquid crystals for understanding of organic electronic devices'

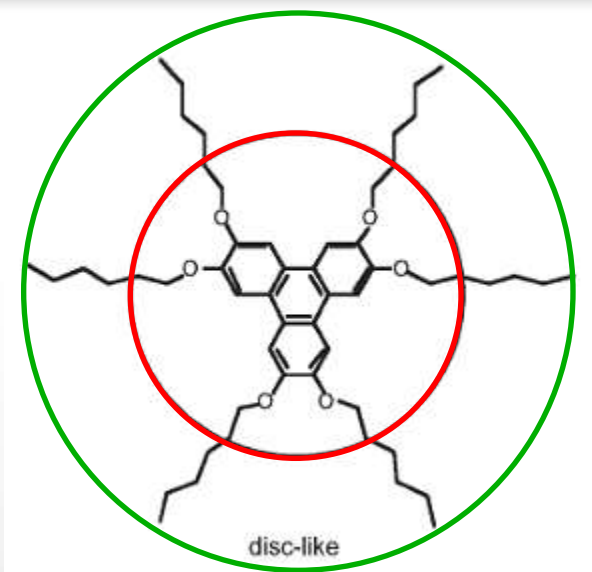


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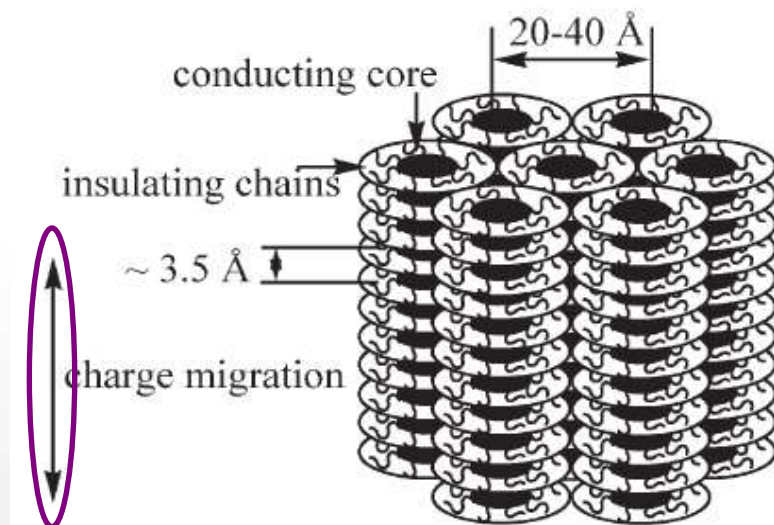
Outline

- Introduction to Discotic Liquid Crystals (DLCs)
- Goal & motivation
- Structure of substrates:
 - gold Au(111)
 - Highly Oriented Pyrolytic Graphite (HOPG)
- Control of DLCs monolayer self-assemblies by molecular engineering
- Summary & perspectives

Introduction to Discotic Liquid Crystals (DLCs)



Shimizu, Y. *et al. J. Mater. Chem.*, **2007**, 17, 4223



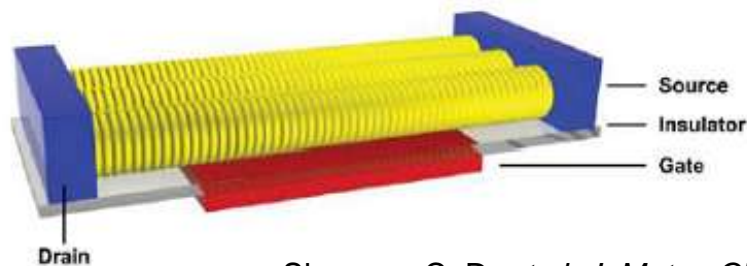
Kumar, S. *Chem. Soc. Rev.* **2006**, 35, 83

2 types of building blocks:

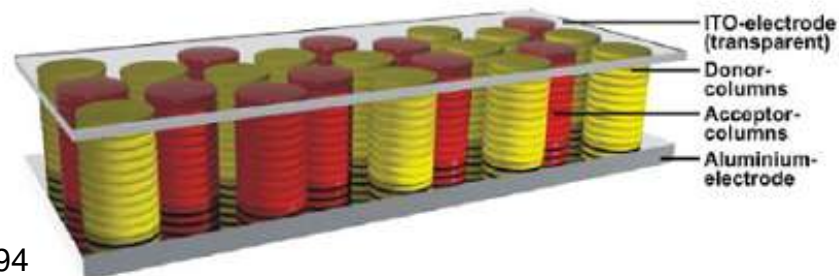
polyaromatic central core

+

multiple peripheral
alkyl or alkoxy tails



VS.

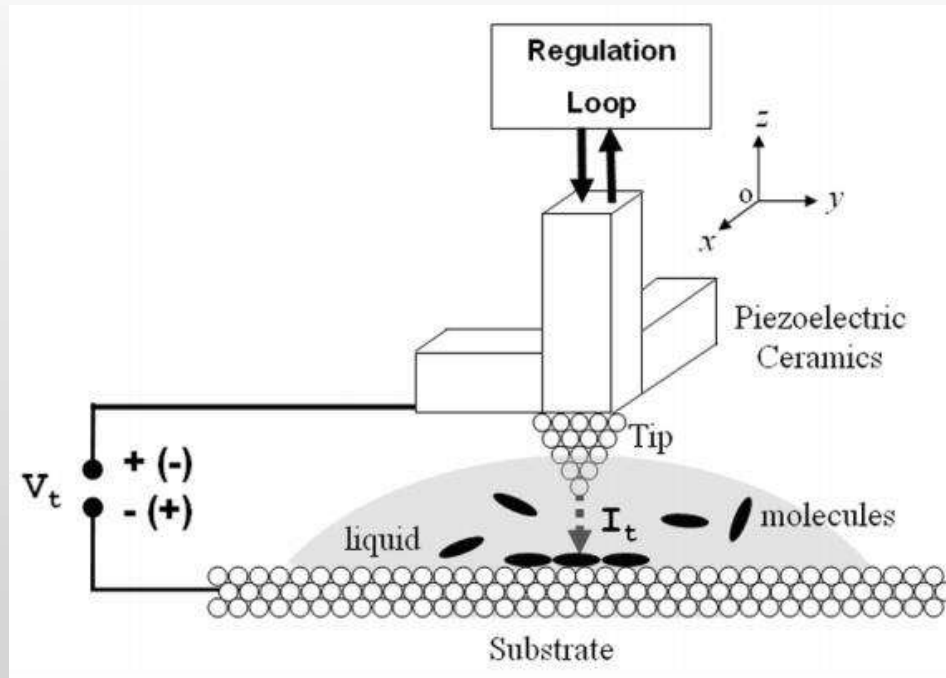


Simpson, C. D. *et al. J. Mater. Chem.*, **2004**, 14, 494

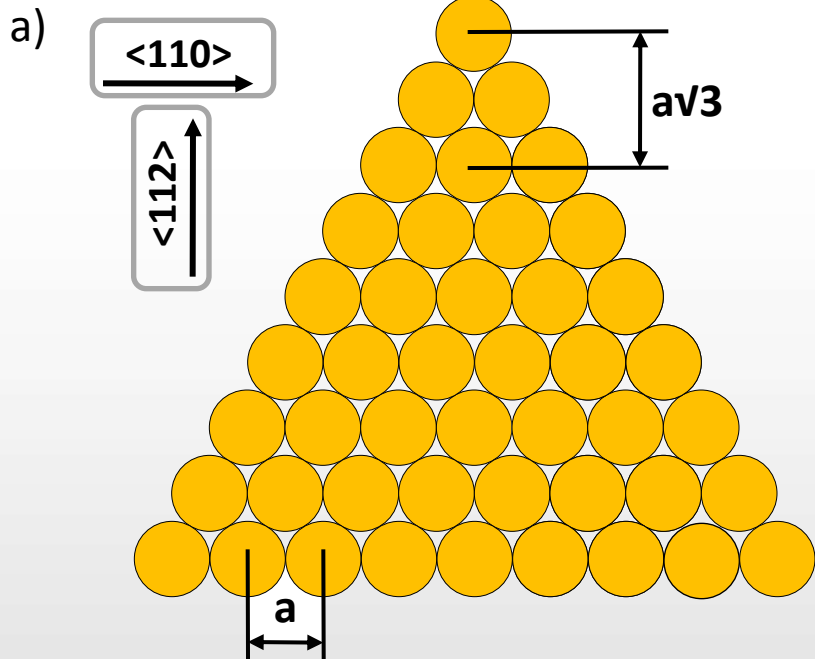
Goal & motivation

- self-assembly of DLCs at crystalline substrates (YES/NO)?
- If YES → resolve structure & understand its origin
- how does the presence of order influence the bulk property (charge transport)?

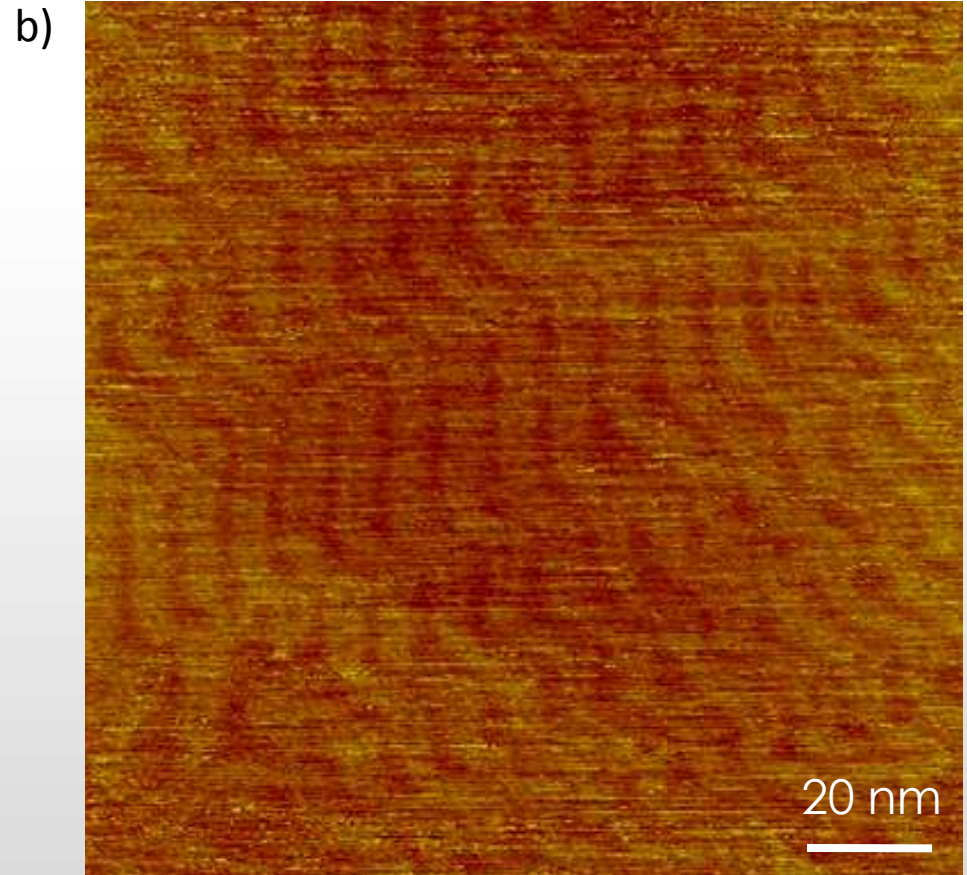
Tool: Scanning Tunneling Microscopy at a liquid/solid interface



Structure of substrates used: Au(111)



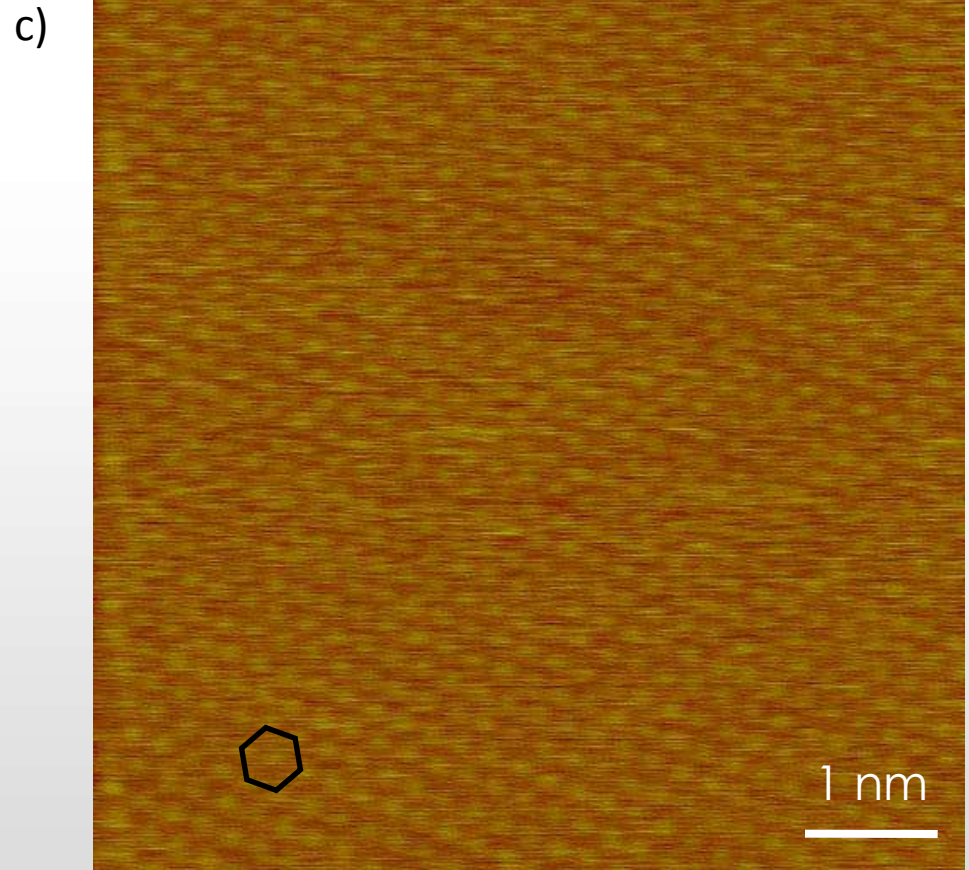
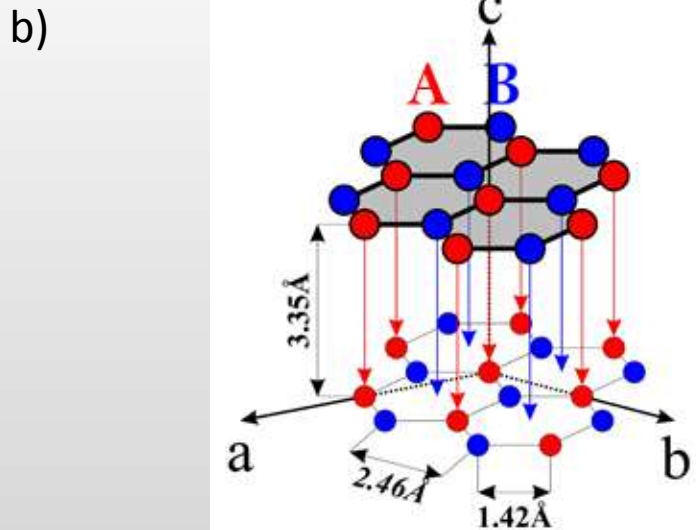
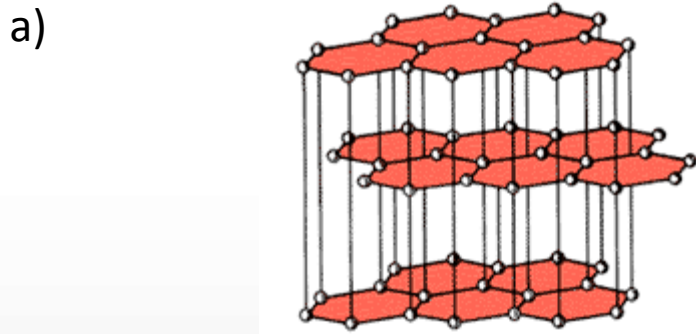
Parameter	Distance [nm]
a	0,288
$a\sqrt{3}$	0,499



bias = -0.200 [V] set point = 1.2 [pA]

- a) hard-sphere representation of non-reconstructed Au(111) surface;
b) STM constant current image of Au(111) reconstructed surface.

Structure of substrates used: graphite



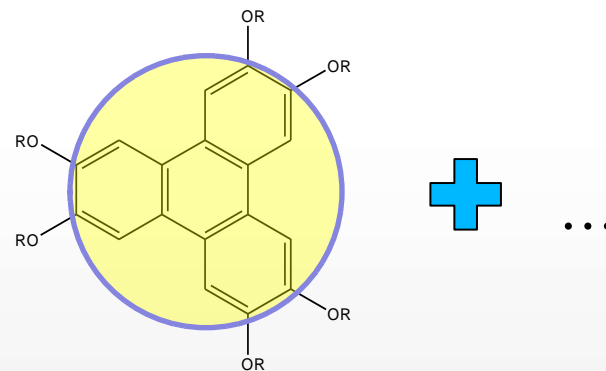
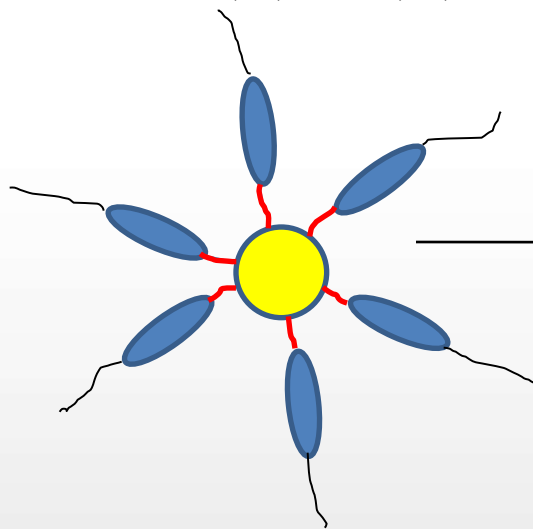
a) molecular structure of graphite; b) scheme indicating shift between two subsequent layers

c) STM constant current image of HOPG

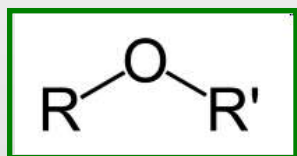
STM of novel photoswitchable mesogenic systems

Synthesis: prof. Kingo Uchida (Ryukoku University, Japan)

Y. Shimizu, Y.; Kurobe, A.; Monobe, H.; Terasawa, N.; Kiyohara, K.; Uchida, K.; *Chem. Commun.*, **2003**, 14, 1676

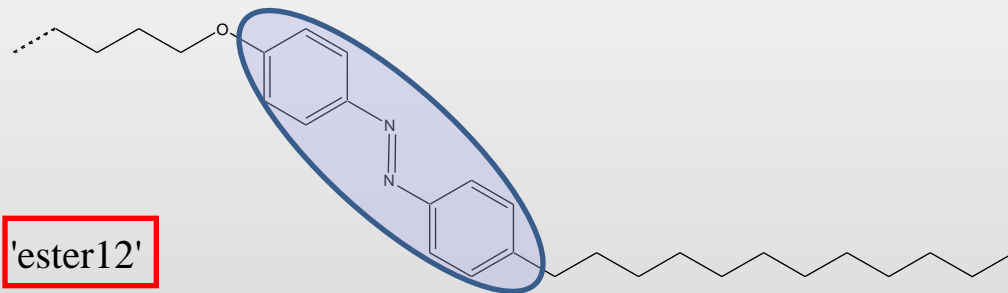


A) ether



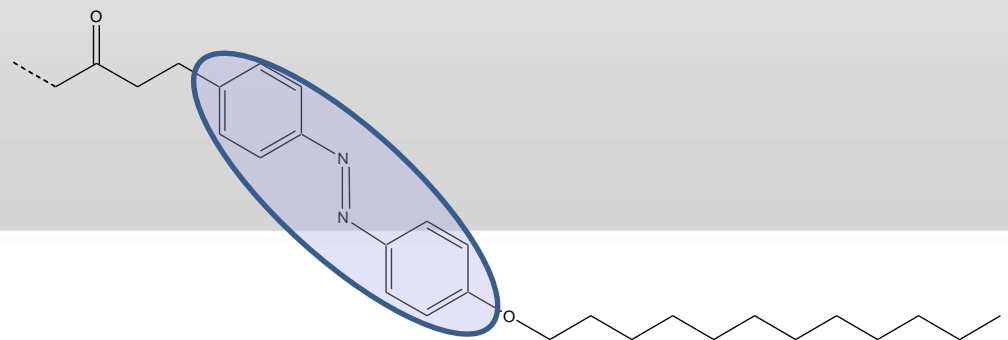
'ether12'

R =

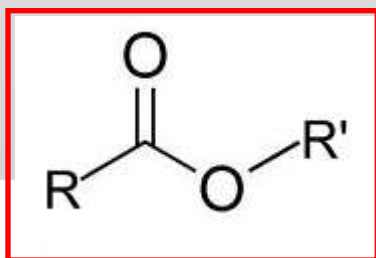


'ester12'

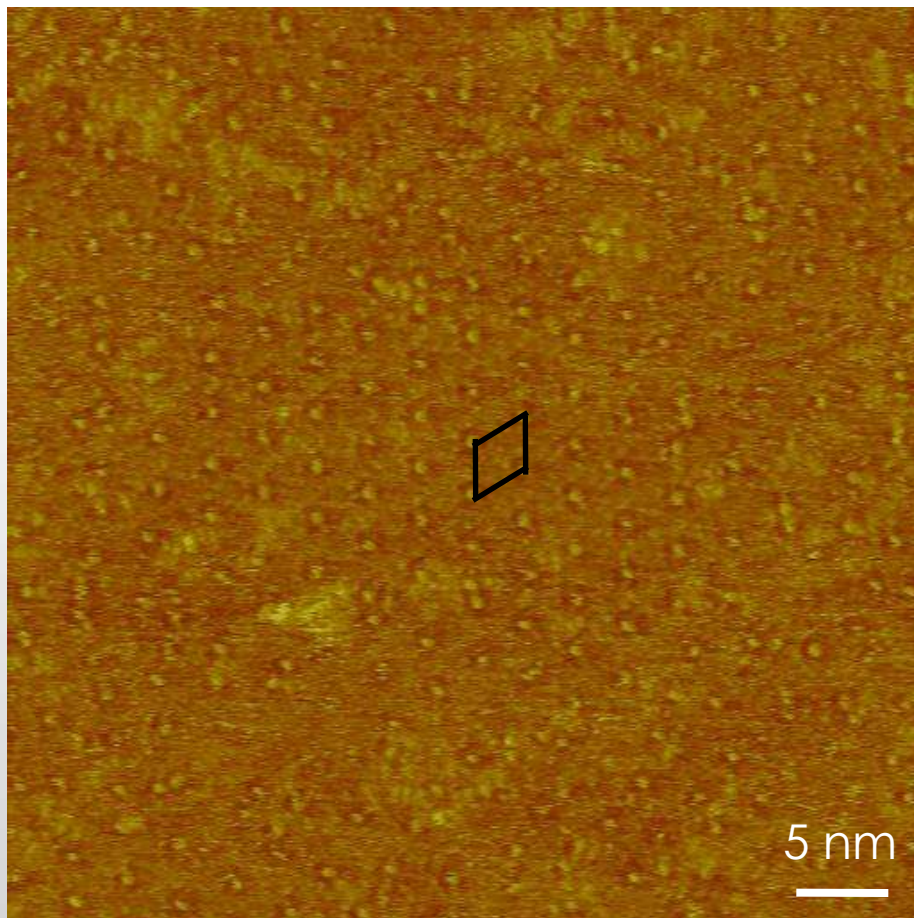
R =



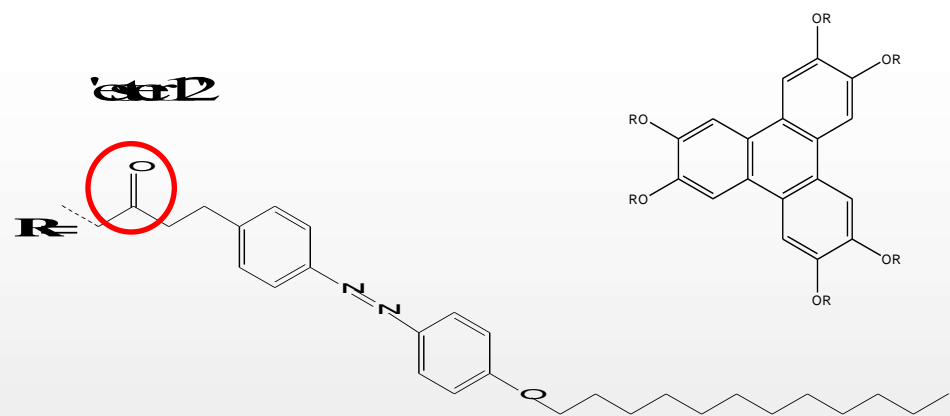
B) ester



Hexagonal symmetry of domains



bias = +0.15 [V] set point = 5 [pA]

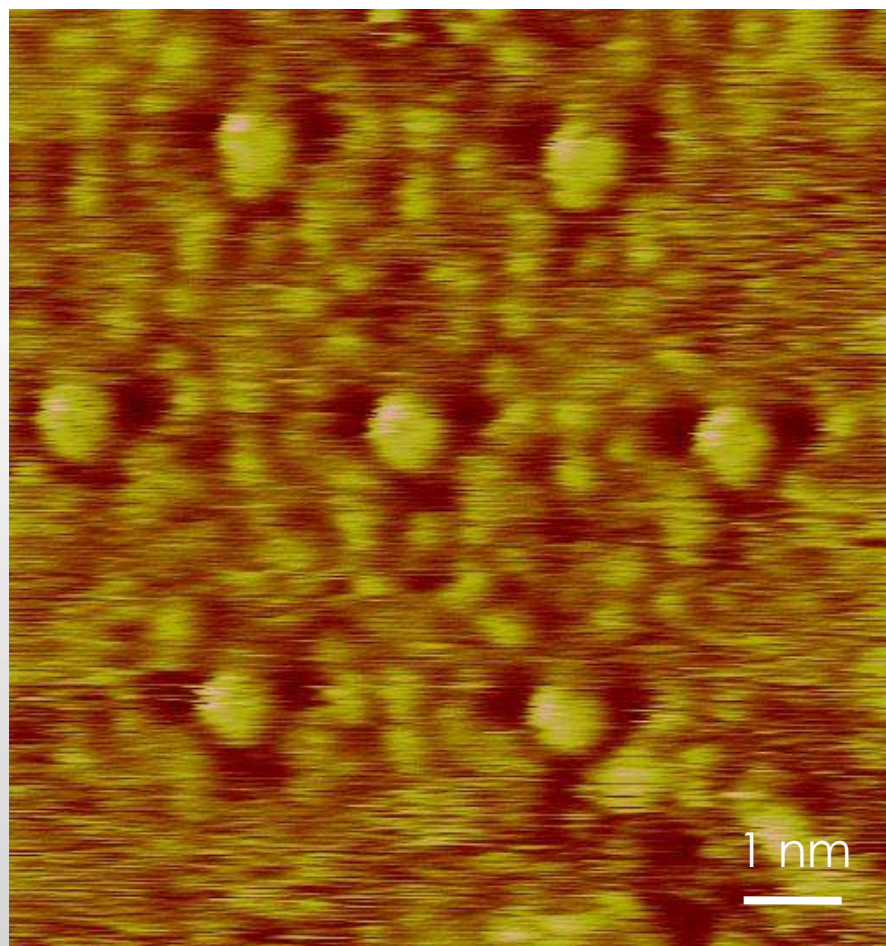


- large areas of coverage
- lattice parameter: **3,5 nm**

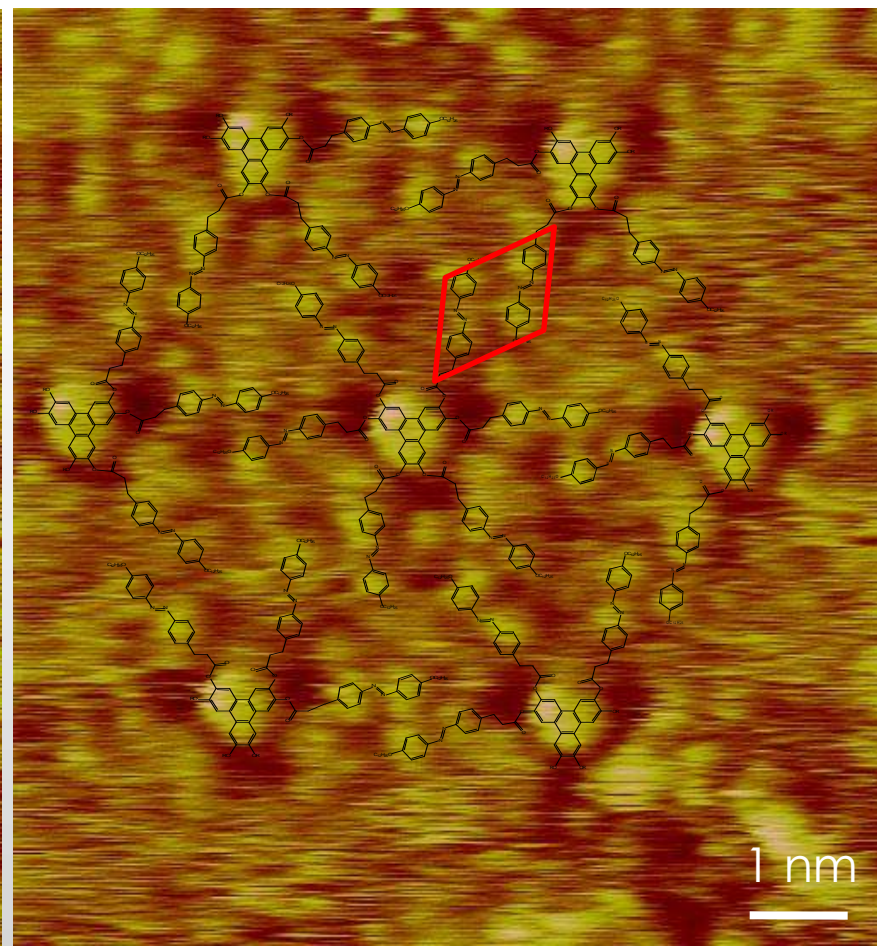
VS

5,66 nm (XRD, Shimizu *et al.*)

Intramolecular resolution of hexagonal-type domain:



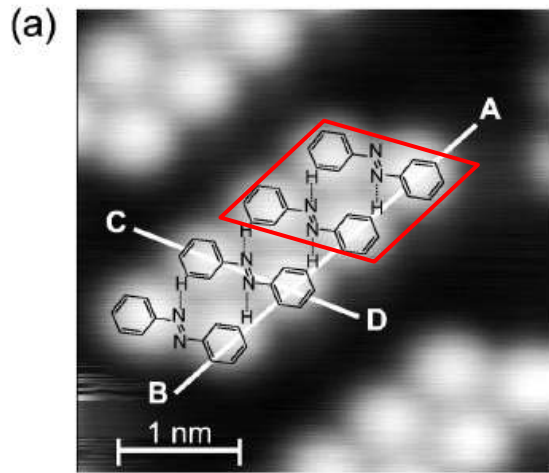
bias = +0.315 [V] set point = 3,93 [pA]



bias = +0.315 [V] set point = 3,93 [pA]

equal interactions with all 6 neighboring molecules, azobenzene-group pairing

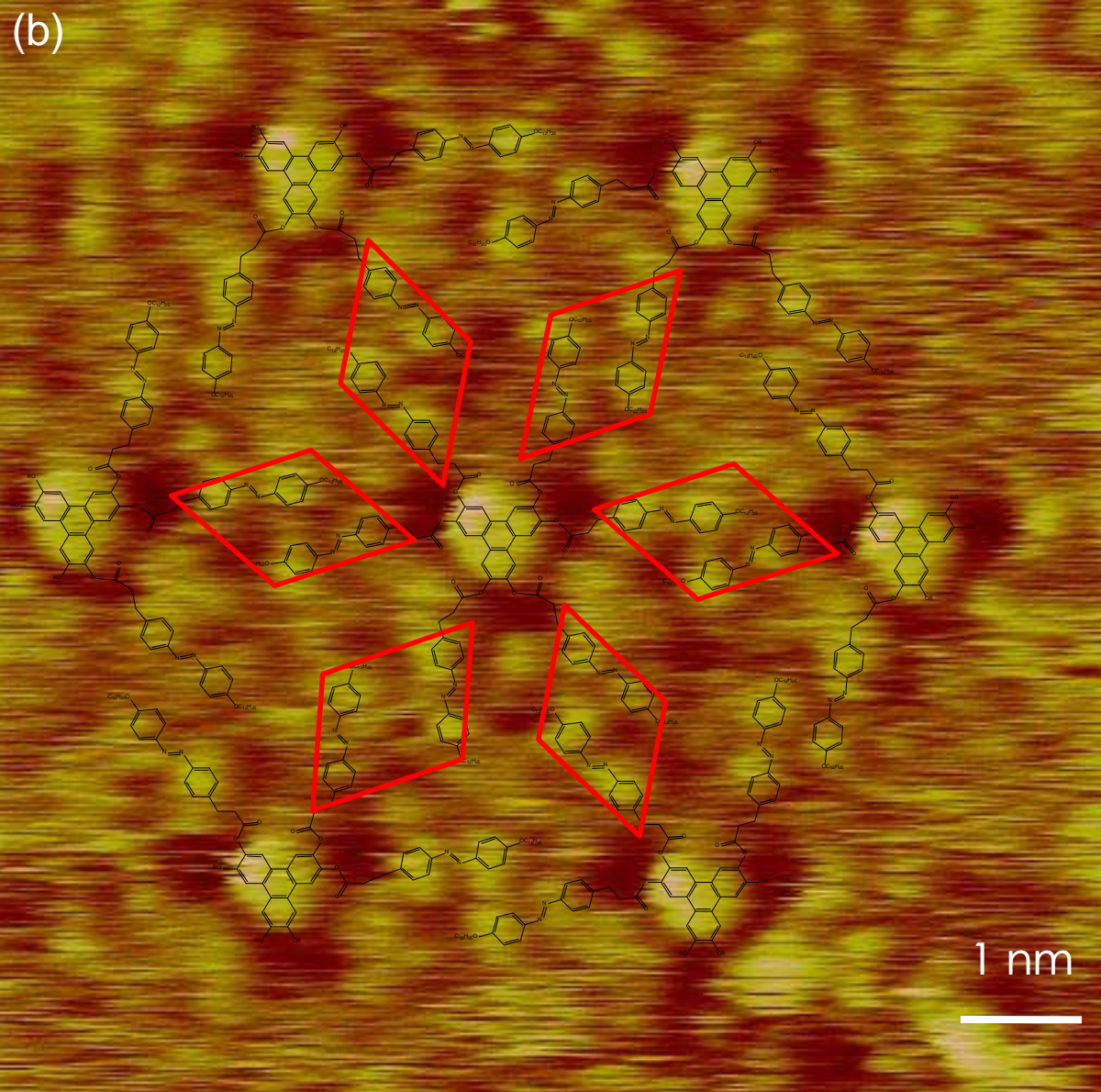
Azobenzene interactions on Au(111)



(a) [Kirakosian, A. *et al.* *Phys. Rev. B*, **2005**, *71*, 113409]

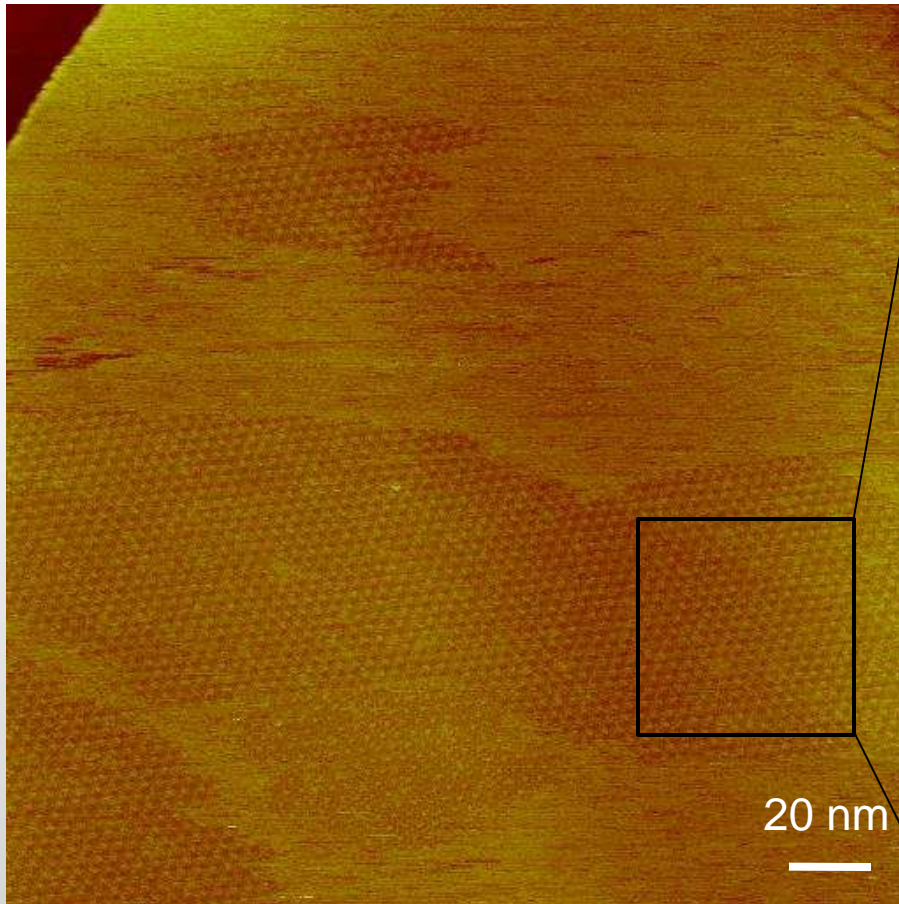
- dipole-dipole interactions
- hydrogenlike bonds

no space for physisorption of alkyl tails!

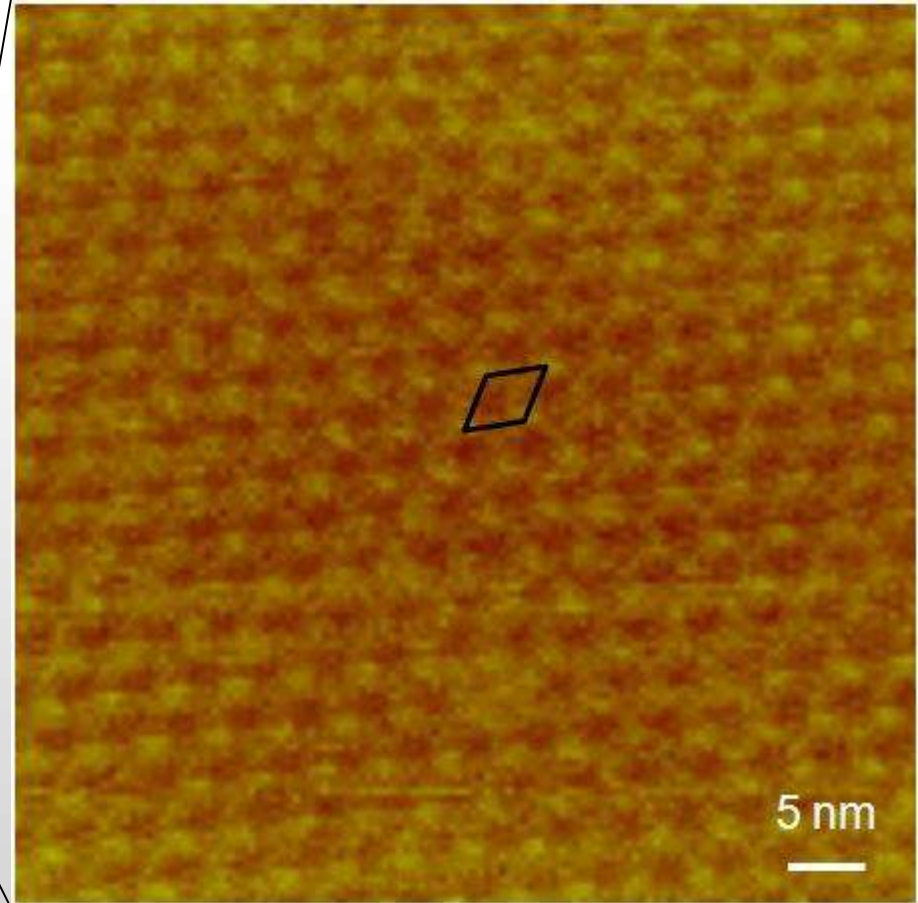


hexagonal network stabilized by intermolecular interactions

Hexagonal packing – preserved



bias = +0.35 [V] set point = 10 [pA]



■ lattice parameter: **4,3 nm**

increased lattice parameter → azobenzene ~~group~~ pairing

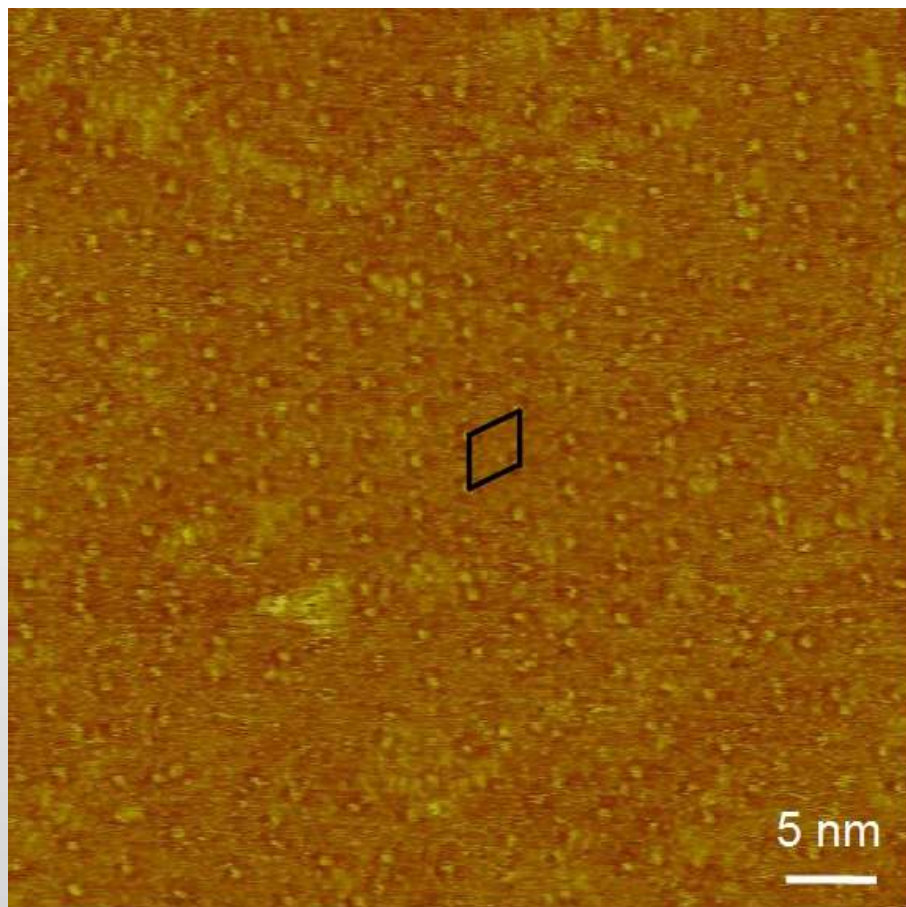
STM of 'ester12': Au(111) vs HOPG comparison

Hexagonal packing – preserved, nearest neighbor distance -

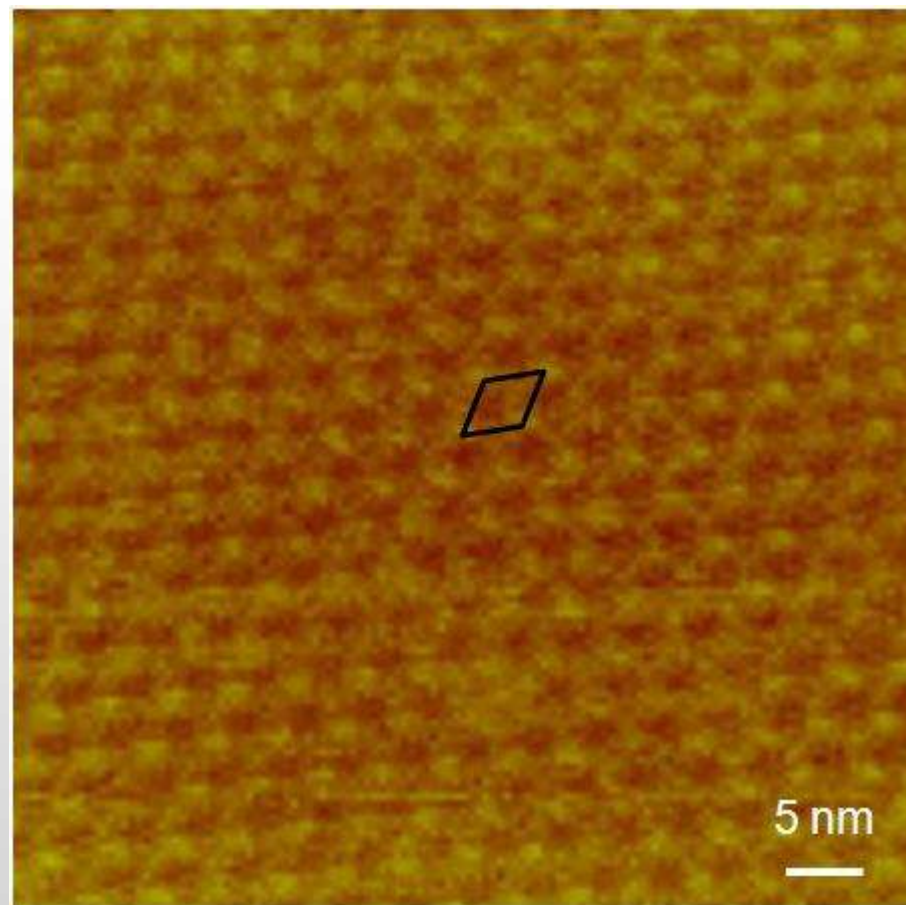
Au(111)

CHANGED

HOPG



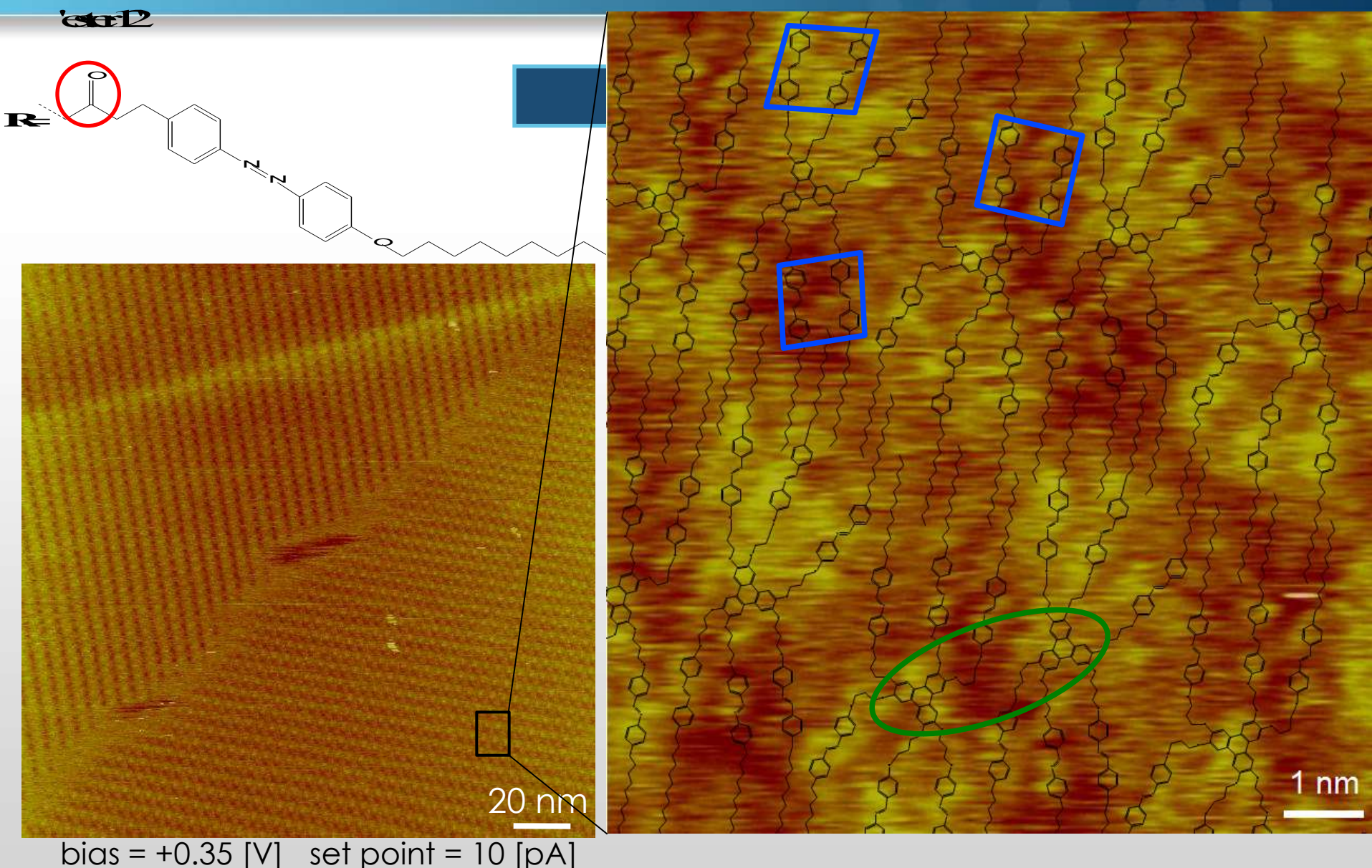
▪ lattice parameter: **3,5 nm**



▪ lattice parameter: **4,3 nm**

Au(111):Y, substrate mediated azobenzene-pairing; HOPG: X

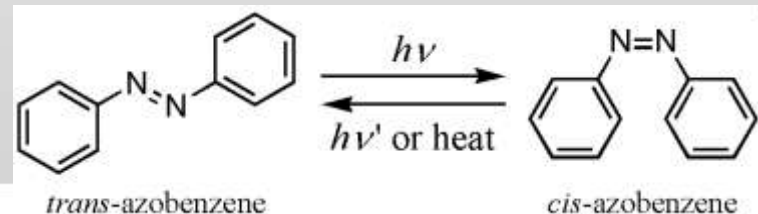
STM of 'ether12' at a 1,2,4-trichlorobenzene /HOPG interface



HOPG: ester to ether → breaking of hexagonal symmetry

Summary & perspectives

- self-assembly of a novel hybrid system at a solid/liquid interface
- **ester**-linked molecule → hexagonal packing (regardless on the substrate type)
 - substrate mediated pairing effect of azobenzene units, on **Au(111)** (stabilizing the monolayer)
- **ester** to **ether** linkage change results in the hexagonal symmetry breaking on HOPG (row-like structure → dimers)
 - **Au(111)**: probably as a result of unfavorable conformational changes of molecule in the solution.
- exploring the switching properties of physisorbed **ester** & **ether** analogues



Acknowledgments



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- Dr Sergey Snegir (Chuiko Institute of Surface Chemistry, NAS of Ukraine)



Thank you for your attention