



UNIVERSITÀ
DEGLI STUDI
DI TORINO
ALMA UNIVERSITAS
TAURINENSIS



Fluorescent hybrid dye-silica nanoparticles: role of the dye structure on the dispersion throughout the silica matrix and overall brightness

Alberto G., Caputo G., Viscardi G., Coluccia S. and Martra G.

*Department of Chemistry
&*

“Nanostructured Interfaces and Surfaces” Centre of Excellence, University of Torino – Italy

Bukovel, August 26 - September 2, 2012

Luminescence &...

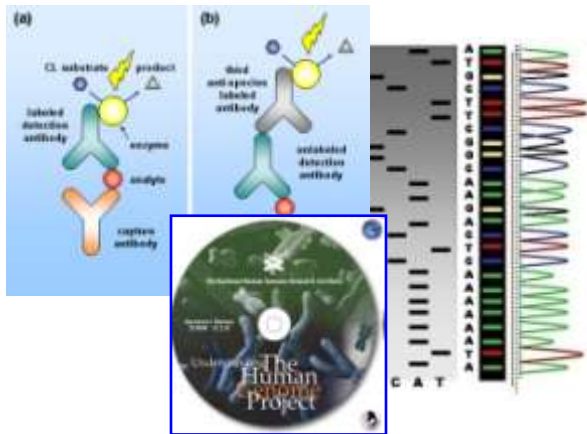


optoelectronics
(often, electrochemiluminescence)

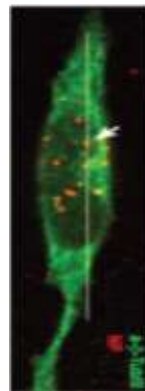


niche applications: forensics
(chemi-, photoluminescence)

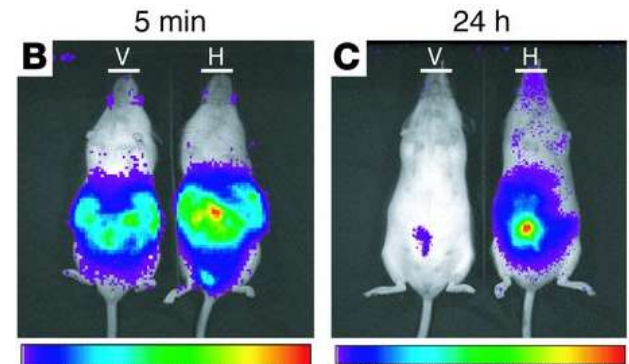
life science (mainly: photoluminescence)



biomolecular



in vitro



in vivo

Photoluminescence, why?

high sensitivity: $C \sim 10^{-12}$ M

ease of excitation (light)

large amount of available photoluminescent compound

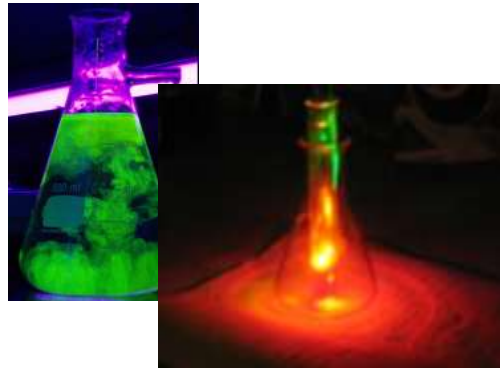
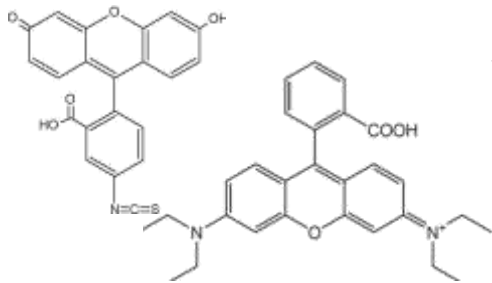
Photoluminescence, why?

high sensitivity: $C \sim 10^{-12}$ M

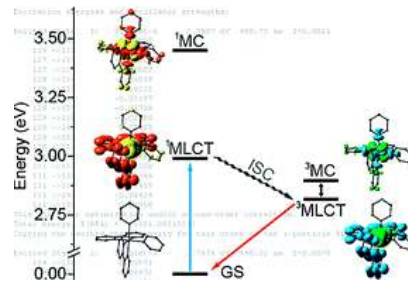
ease of excitation (light)

large amount of available photoluminescent compound

Organic fluorescent dyes



Organometallic complexes



Proteins



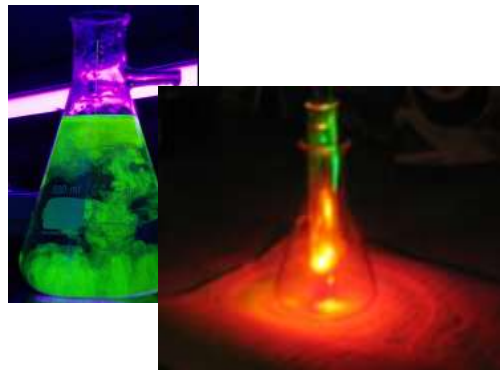
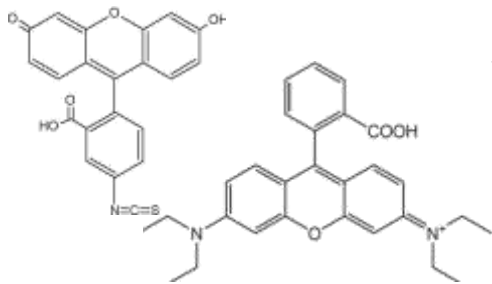
Photoluminescence, why?

high sensitivity: $C \sim 10^{-12}$ M

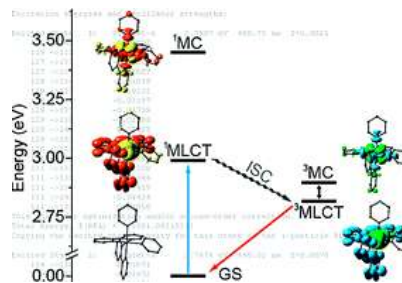
ease of excitation (light)

large amount of available photoluminescent compound

Organic fluorescent dyes



Organometallic complexes

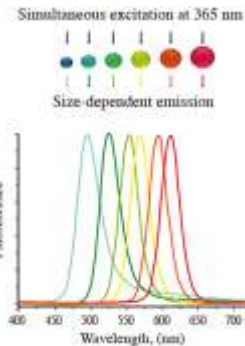


Proteins

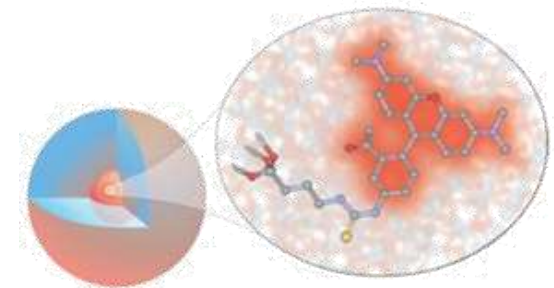


Photoluminescent nanomaterials

Quantum Dots

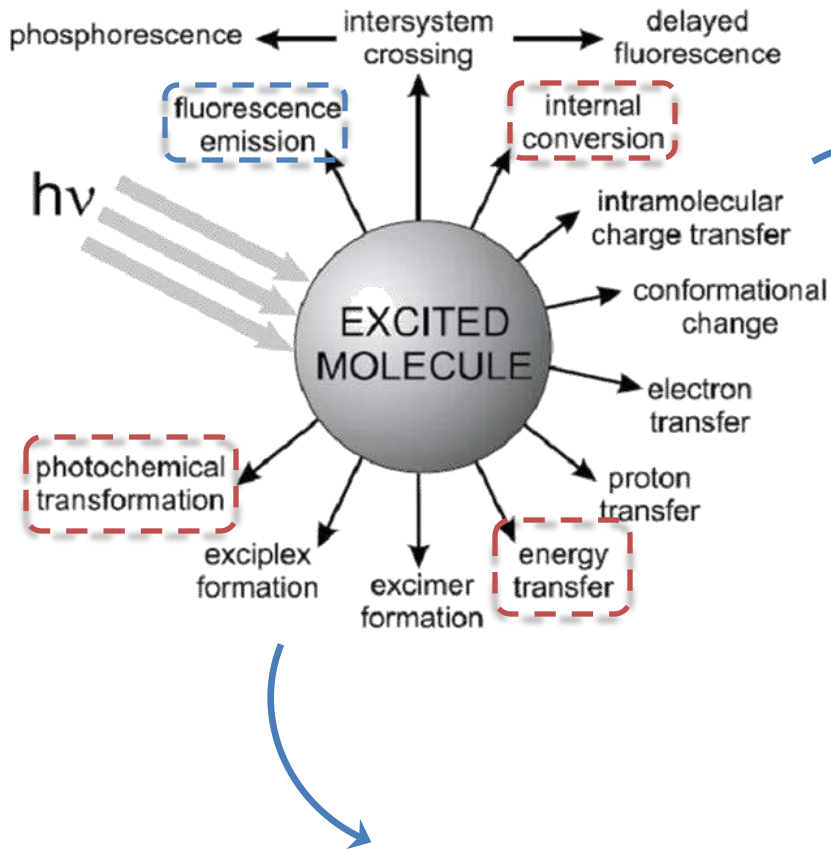


Hybrid organic dye/silica nanoparticles



An emerging tool: Hybrid organic dye/silica nanoparticles

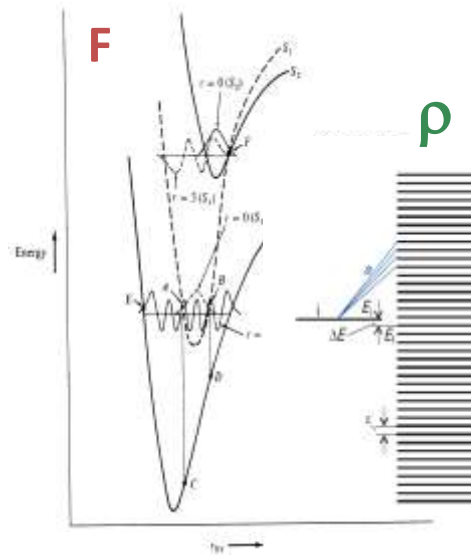
main competitors



prevention of the interaction with O_2

Higher stability

quantum mechanics can help



$$k_{nr} = 2\pi\hbar^{-1} \beta^2 \rho F$$

also the rotation levels must be taken into account

embedding of molecules decreases ρ

Increase of molecular rigidity

Enhanced brightness

Influence of the particle architecture on the photophysical properties

Base-catalyzed hydrolysis and polycondensation of silica precursors and organic dyes-silane derivatives in:

- a) Homogeneous solutions (Stöber method)
- b) Reverse microemulsion

Influence of the particle architecture on the photophysical properties

Base-catalyzed hydrolysis and polycondensation of silica precursors and organic dyes-silane derivatives in:

- a) Homogeneous solutions (Stöber method)
- b) Reverse microemulsion

The performance of the final material is strongly determined by the distribution of the fluorophore molecules throughout the host matrix

Influence of the particle architecture on the photophysical properties

Base-catalyzed hydrolysis and polycondensation of silica precursors and organic dyes-silane derivatives in:

- Homogeneous solutions (Stöber method)
- Reverse microemulsion

The performance of the final material is strongly determined by the distribution of the fluorophore molecules throughout the host matrix

CHEMISTRY OF
MATERIALS

Subscriber access provided by UNIV STUDI DI TORINO

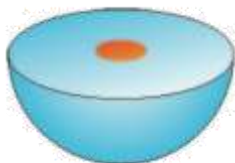
Article

Silica Nanoparticle Architecture Determines Radiative Properties of Encapsulated Fluorophores

Daniel R. Larson, Hooisweng Ow, Harshad D. Vishwasrao, Ahmed A. Heikal, Ulrich Wiesner, and Watt W. Webb

Chem. Mater., 2008, 20 (8), 2677-2684 • DOI: 10.1021/cm7026866 • Publication Date (Web): 15 March 2008

Downloaded from <http://pubs.acs.org> on November 17, 2008



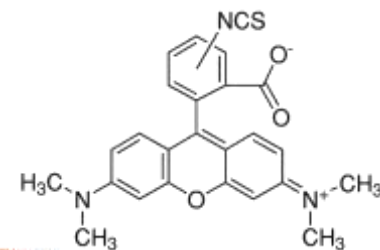
Compact core-shell



Expanded core-shell



Homogeneous particles



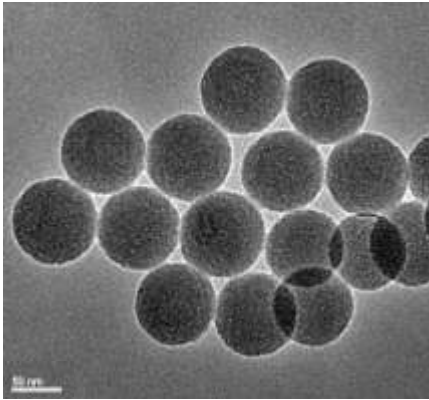
TRITC

sample	$\langle \tau_f \rangle$ (ns)	φ	k_r (ns ⁻¹)	k_r normalized	k_{nr} (ns ⁻¹)	k_{nr} normalized	$\langle \theta \rangle$ (ns)
TRITC	2.1	0.15 ^b	0.072	1.0	0.41	1.0	0.21
compact core-shell	1.8	0.30	0.17	2.3	0.39	0.95	0.40
expanded core-shell	2.9	0.47	0.16	2.2	0.18	0.45	3.1
homogeneous	3.2	0.50	0.16	2.2	0.16	0.39	5.9

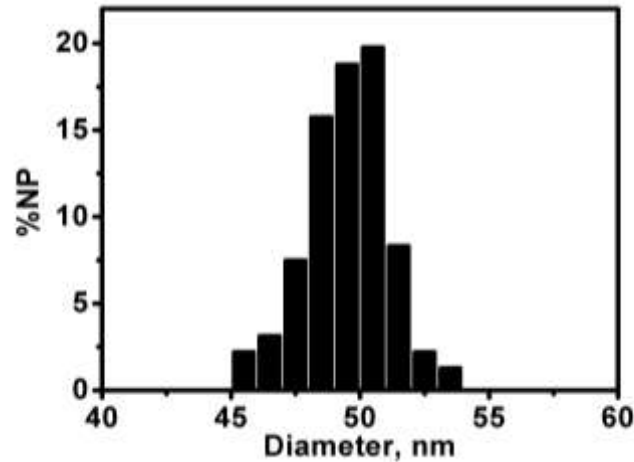
homogeneous fluorophore distribution provides best photophysical performances

This Work

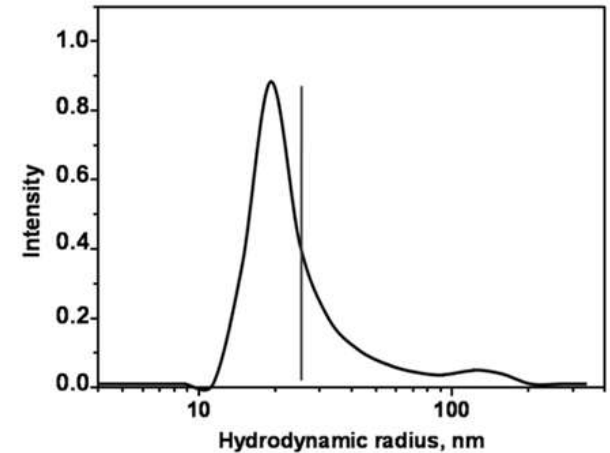
Hybrid dye-SiO₂ NPs prepared by TEOS hydrolysis and polymerization in reverse micelles in
W/O microemulsion



homogeneous spherical
morphology



size monodispersity



good dispersion in water
(pH 5.5)

homogeneous behaviour in functional tests

statistical significance of quantitative evaluation

number of NPs per sample

number of Dye molecules per NP

Possibility to quantitatively evaluate the
photophysical behaviour of entrapped fluorophores

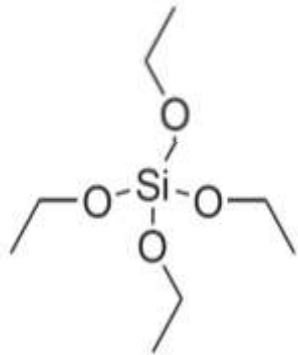
Reverse Microemulsion Technique

Key point:

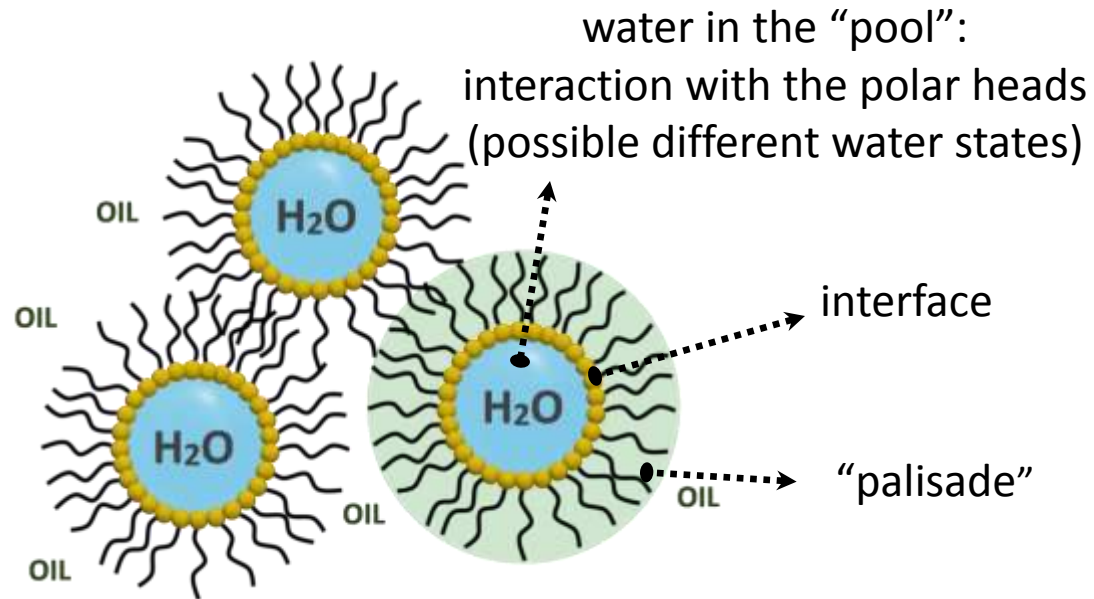
partition of reactants between oil and water, mediated by the micellar “palisade”

Reactants:

Tetraethylorthosilicate (TEOS)



Dye-silane derivative



surfactant:
Triton X-100, non-ionic

co-surfactant:
n-hexanol

oil:
cyclohexane

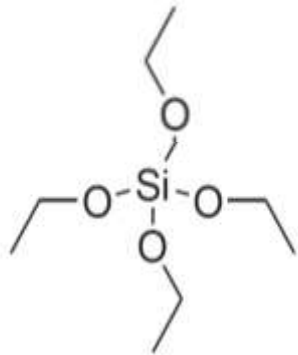
Reverse Microemulsion Technique

Key point:

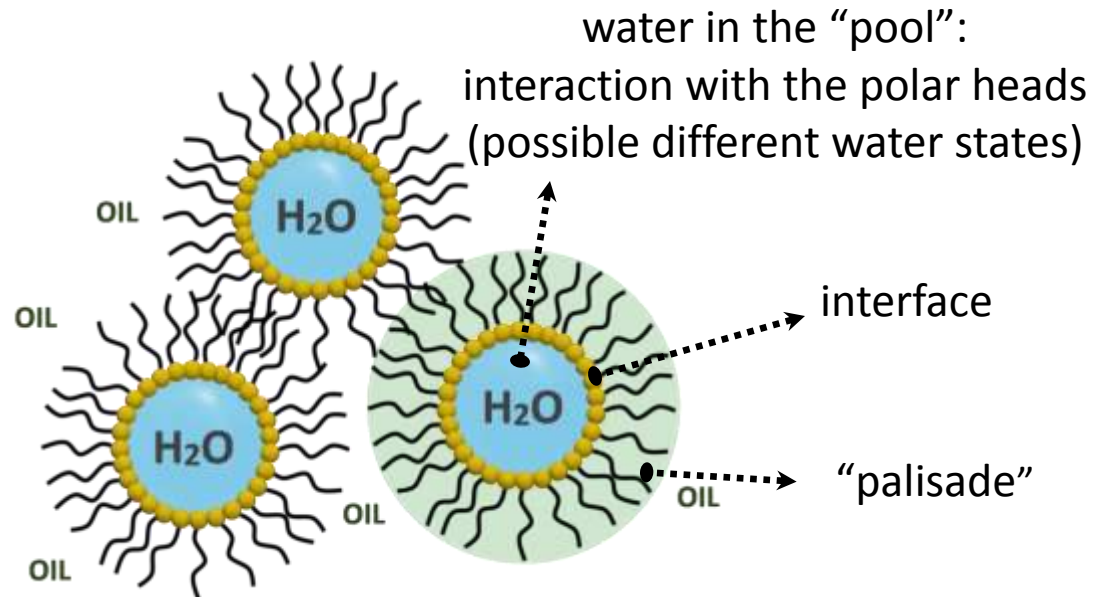
partition of reactants between oil and water, mediated by the micellar “palisade”

Reactants:

Tetraethylorthosilicate (TEOS)



Dye-silane derivative



surfactant:
Triton X-100, non-ionic

co-surfactant:
n-hexanol

oil:
cyclohexane

Goal:

identification of a molecular parameter that could effectively help to tailor the dispersion of the dye molecules in the final hybrid NPs

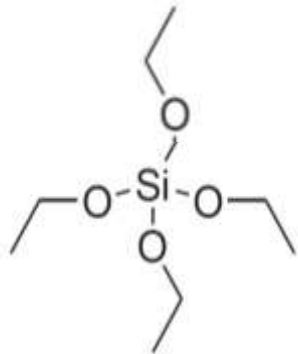
Reverse Microemulsion Technique

Key point:

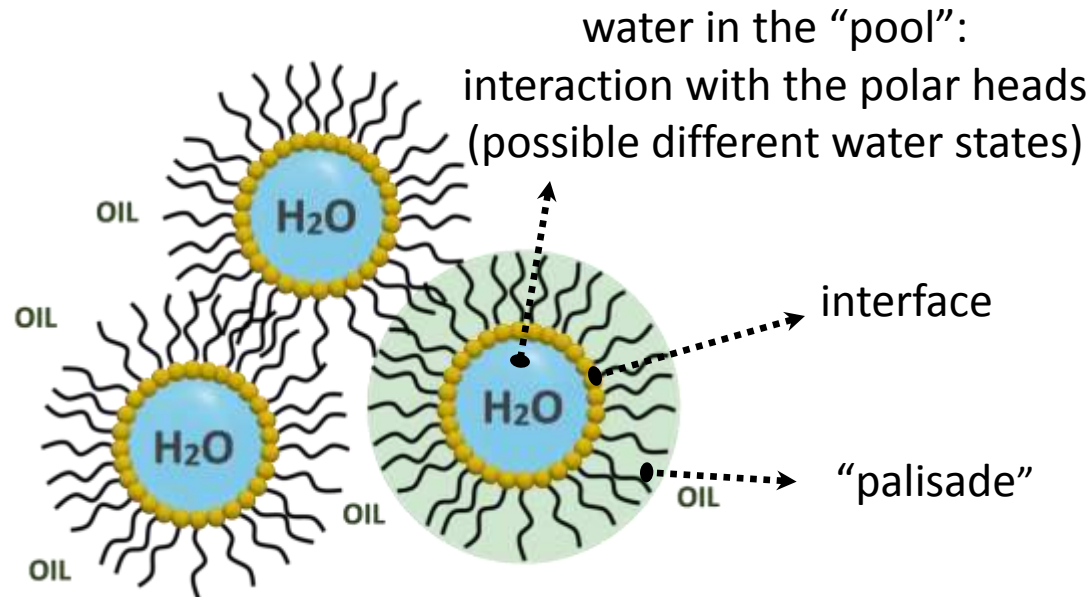
partition of reactants between oil and water, mediated by the micellar “palisade”

Reactants:

Tetraethylorthosilicate (TEOS)



Dye-silane derivative



surfactant:
Triton X-100, non-ionic

co-surfactant:
n-hexanol

oil:
cyclohexane

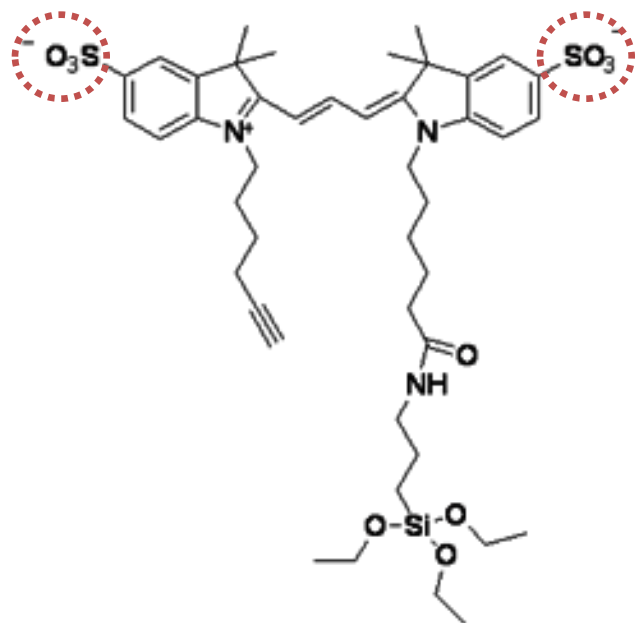
Goal:

identification of a molecular parameter that could effectively help to tailor the dispersion of the dye molecules in the final hybrid NPs

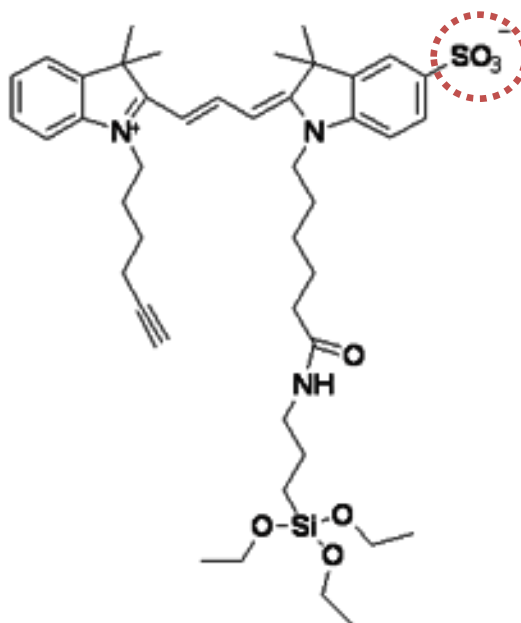
Molecular parameter selected:

relative hydrophobicity of TEOS and Dyes

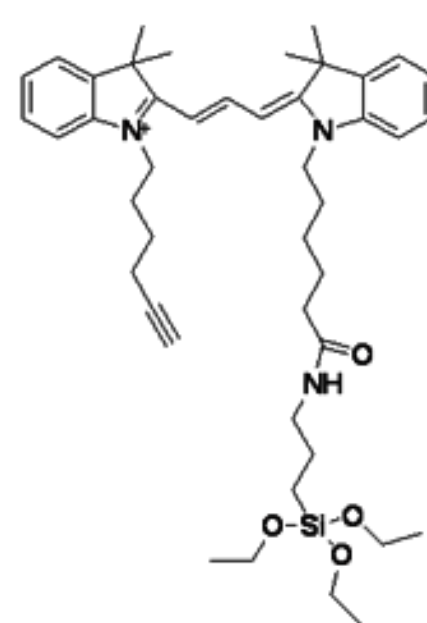
Fluorescent Dyes: Cyanines



I3BS-APTS



I3MS-APTS

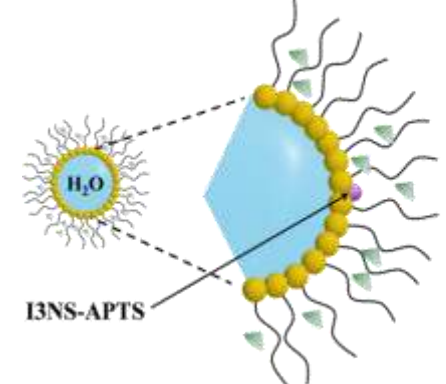
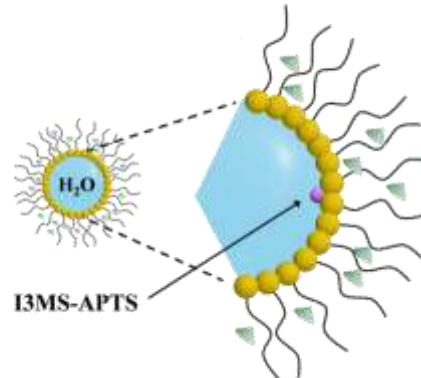
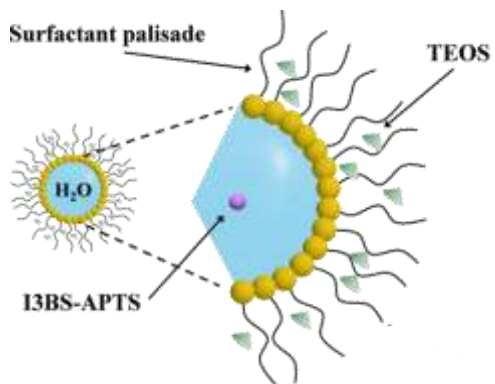
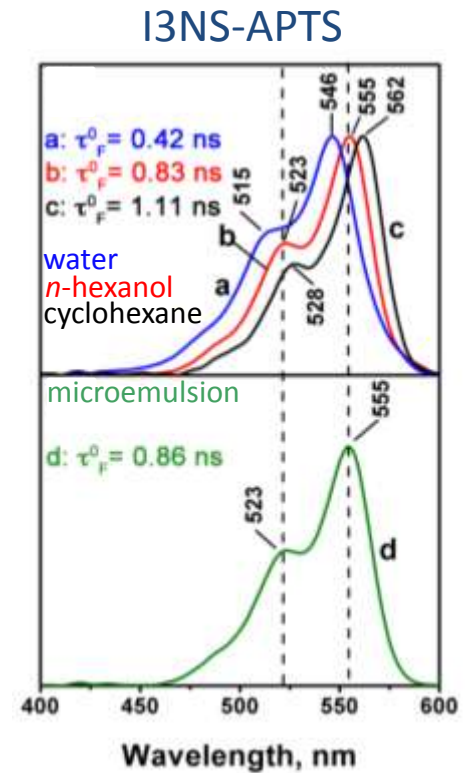
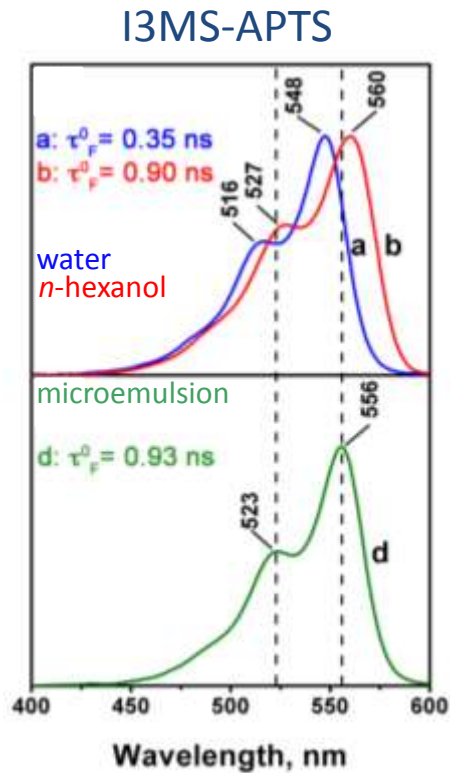
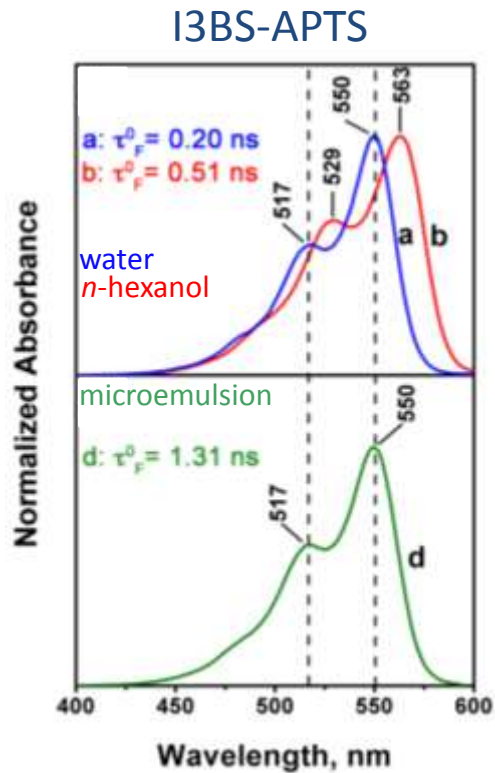


I3NS-APTS



Expected different partition between oil/reverse micelle palisade/water pool

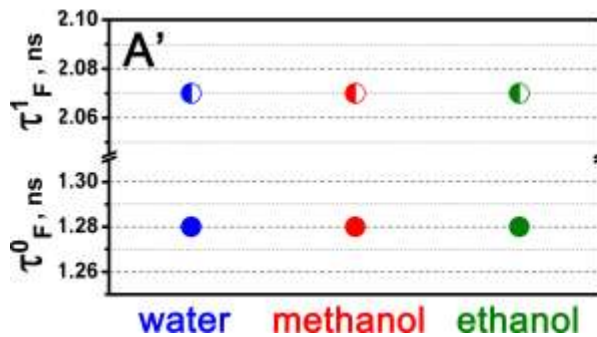
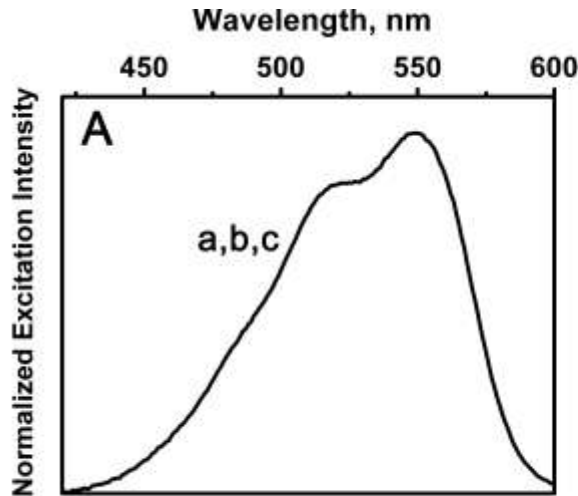
Cyanines at the starting-line



different location of Cy with respect the oil/water interface depending on their hydrophobicity

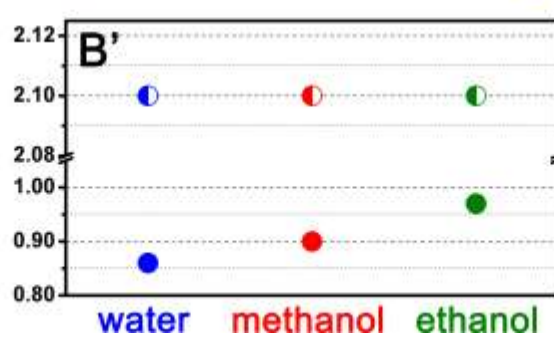
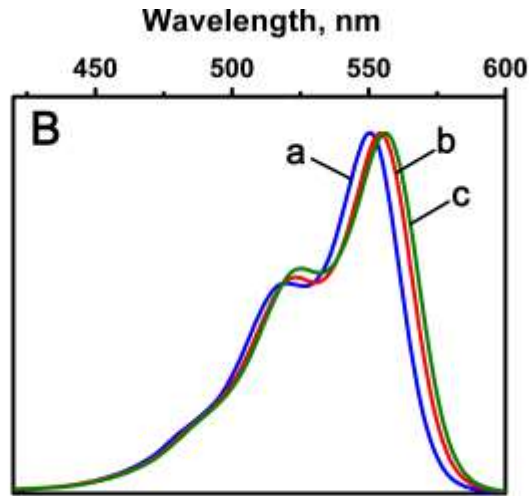
Cyanines location in the final materials

I3BS-NPs



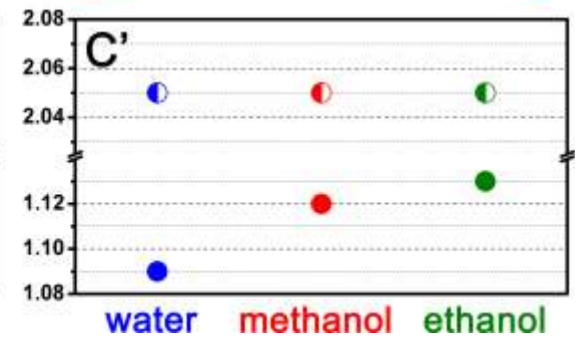
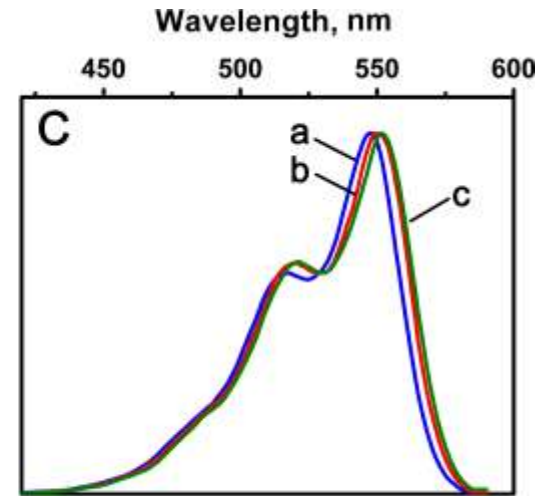
Complete inclusion of I3BS-APTS molecules within the NPs

I3MS-NPs



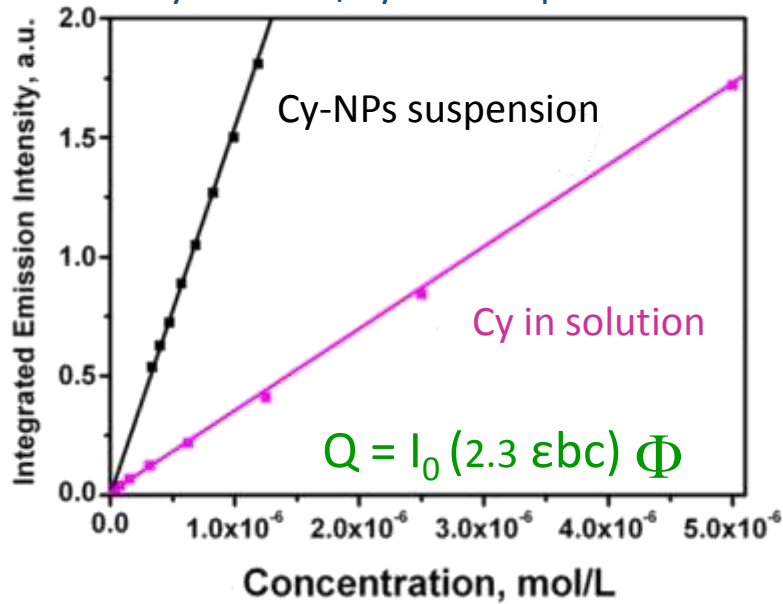
Presence of some I3MS-APTS molecules on the surface

I3NS-NPs



Presence of some I3NS-APTS molecules on the surface

Photoluminescence intensity of equimolar
Cy solution/Cy-NPs suspensions



A) Photoluminescence intensity, n. NPs/mL and n. Cy/NP allow to determine the

average gain in Φ

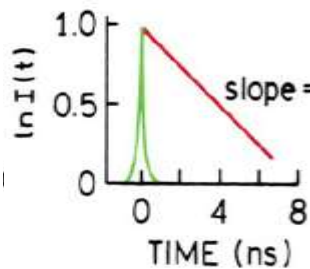
per Cy molecule entrapped in a NP
averaged on **ALL** Cy molecules in a NP
(both fluorescent and not-fluorescent)

B) fluorescence lifetime ($\tau \propto \Phi$):

-signal **ONLY** due to fluorescent Cy

- gain in τ (Φ) dependent on the Cy environment

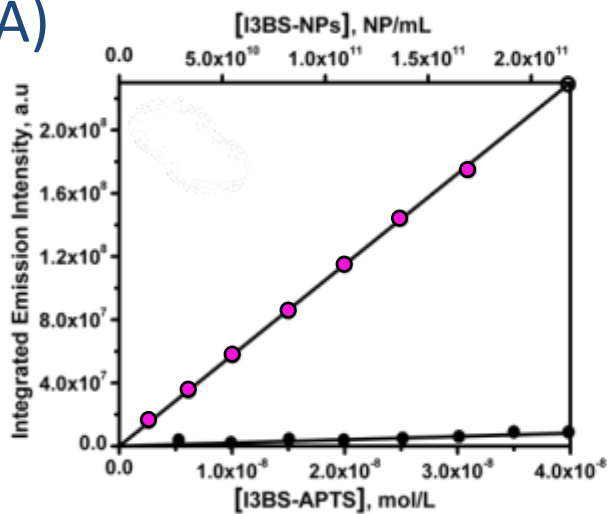
number and relative extent of “families” of fluorescent Cy



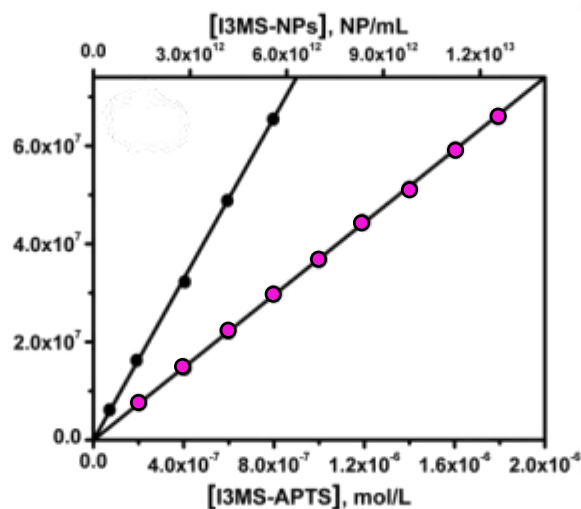
τ^0_F (ns)	1.09 (35%)
τ^1_F (ns)	2.05 (65%)
χ^2	1.09

quantitative treatment of photoluminescent intensity and lifetime data

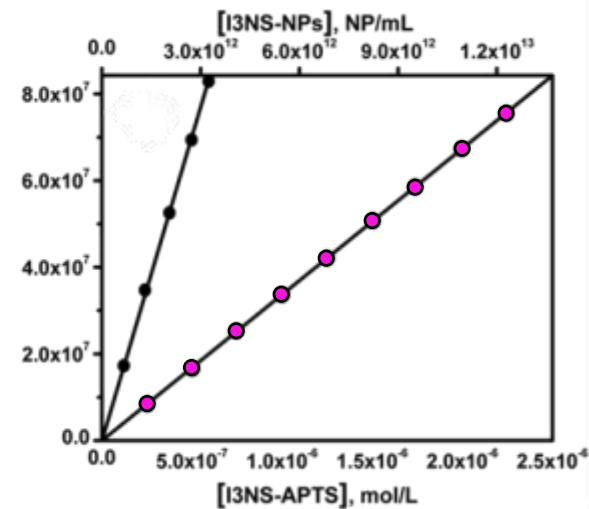
A)



$$\Phi_{\text{NPs}} / \Phi_{\text{Cy}} : 0.04$$



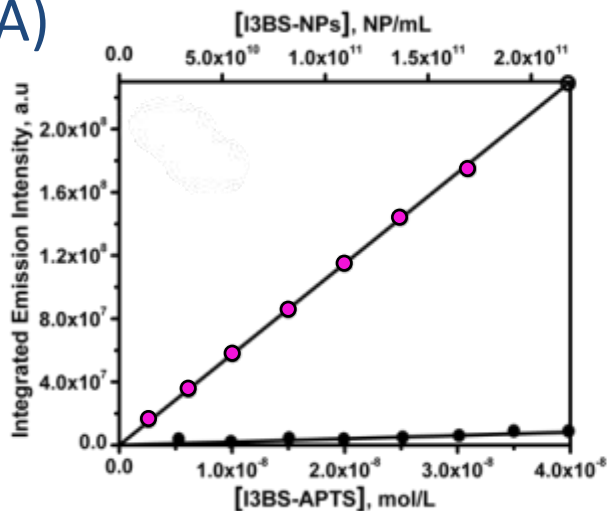
$$\Phi_{\text{NPs}} / \Phi_{\text{Cy}} : 2.7$$



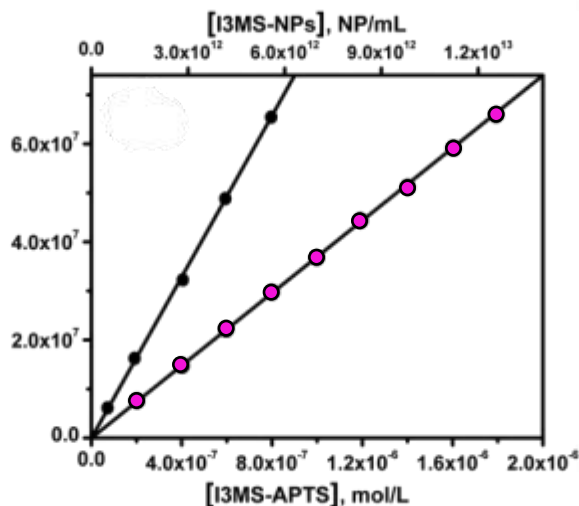
$$\Phi_{\text{NPs}} / \Phi_{\text{Cy}} : 4.3$$

quantitative treatment of photoluminescent intensity and lifetime data

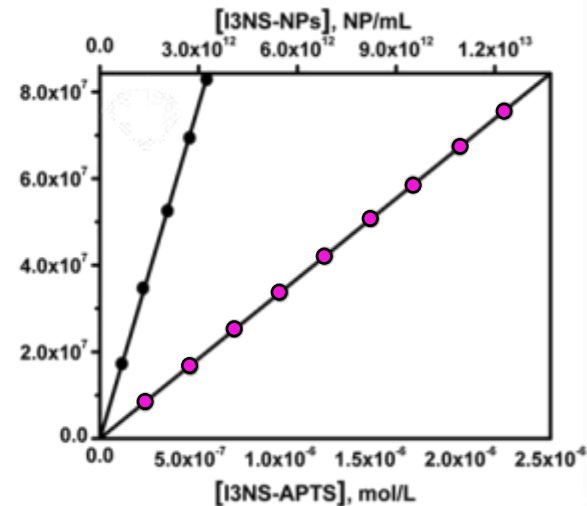
A)



$$\Phi_{\text{NPs}} / \Phi_{\text{Cy}} : 0.04$$



$$\Phi_{\text{NPs}} / \Phi_{\text{Cy}} : 2.7$$



$$\Phi_{\text{NPs}} / \Phi_{\text{Cy}} : 4.3$$

B)

Relative Φ expected gain: $\sum x_i(\tau_i \text{NP} / \tau_i \text{Mol})$
(averaged on EMITTING Cy)

	I3DS-APTS	I3DS-NPs
τ_F^0	0.20 (100%)	1.28 (55%)
τ_F^1	---	2.06 (45%)
χ^2	1.02	1.03

5.7

	I3MS-APTS	I3MS-NPs
τ_F^0	0.35 (100%)	0.86 (47%)
τ_F^1	---	2.10 (53%)
χ^2	1.09	1.09

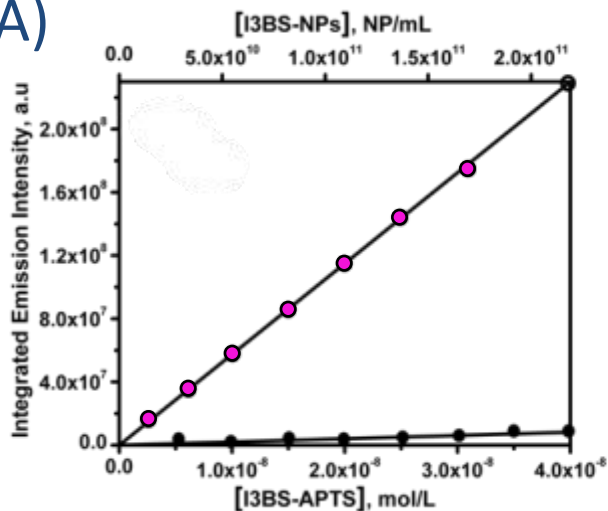
4.3

	I3NS-APTS	I3NS-NPs
τ_F^0	0.42 (100%)	1.09 (35%)
τ_F^1	---	2.05 (65%)
χ^2	1.2	1.09

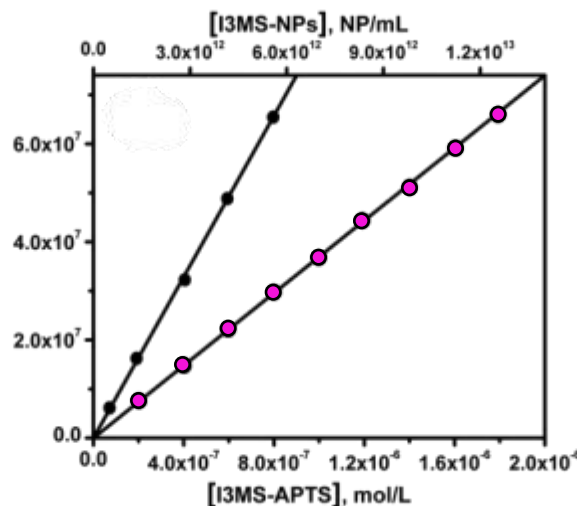
4.2

quantitative treatment of photoluminescent intensity and lifetime data

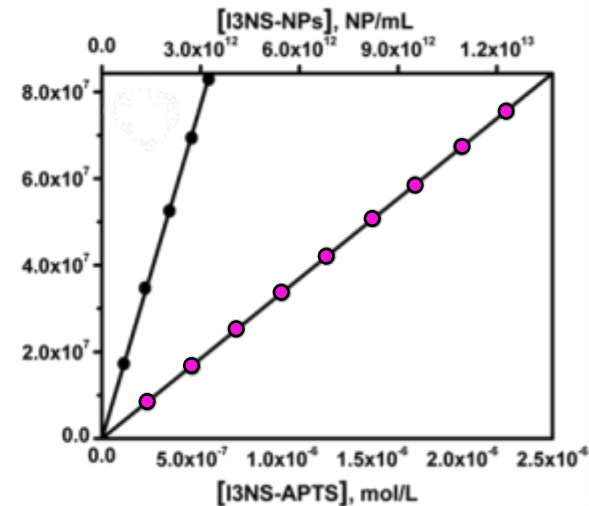
A)



$$\Phi_{\text{NPs}} / \Phi_{\text{Cy}} : 0.04$$



$$\Phi_{\text{NPs}} / \Phi_{\text{Cy}} : 2.7$$



$$\Phi_{\text{NPs}} / \Phi_{\text{Cy}} : 4.3$$

B)

Relative Φ expected gain: $\sum x_i(\tau_i \text{NP} / \tau_i \text{Mol})$
(averaged on EMITTING Cy)

	I3DS-APTS	I3DS-NPs
τ_F^0	0.20 (100%)	1.28 (55%)
τ_F^1	---	2.06 (45%)
χ^2	1.02	1.03

5.7

	I3MS-APTS	I3MS-NPs
τ_F^0	0.35 (100%)	0.86 (47%)
τ_F^1	---	2.10 (53%)
χ^2	1.09	1.09

4.3

	I3NS-APTS	I3NS-NPs
τ_F^0	0.42 (100%)	1.09 (35%)
τ_F^1	---	2.05 (65%)
χ^2	1.2	1.09

4.2

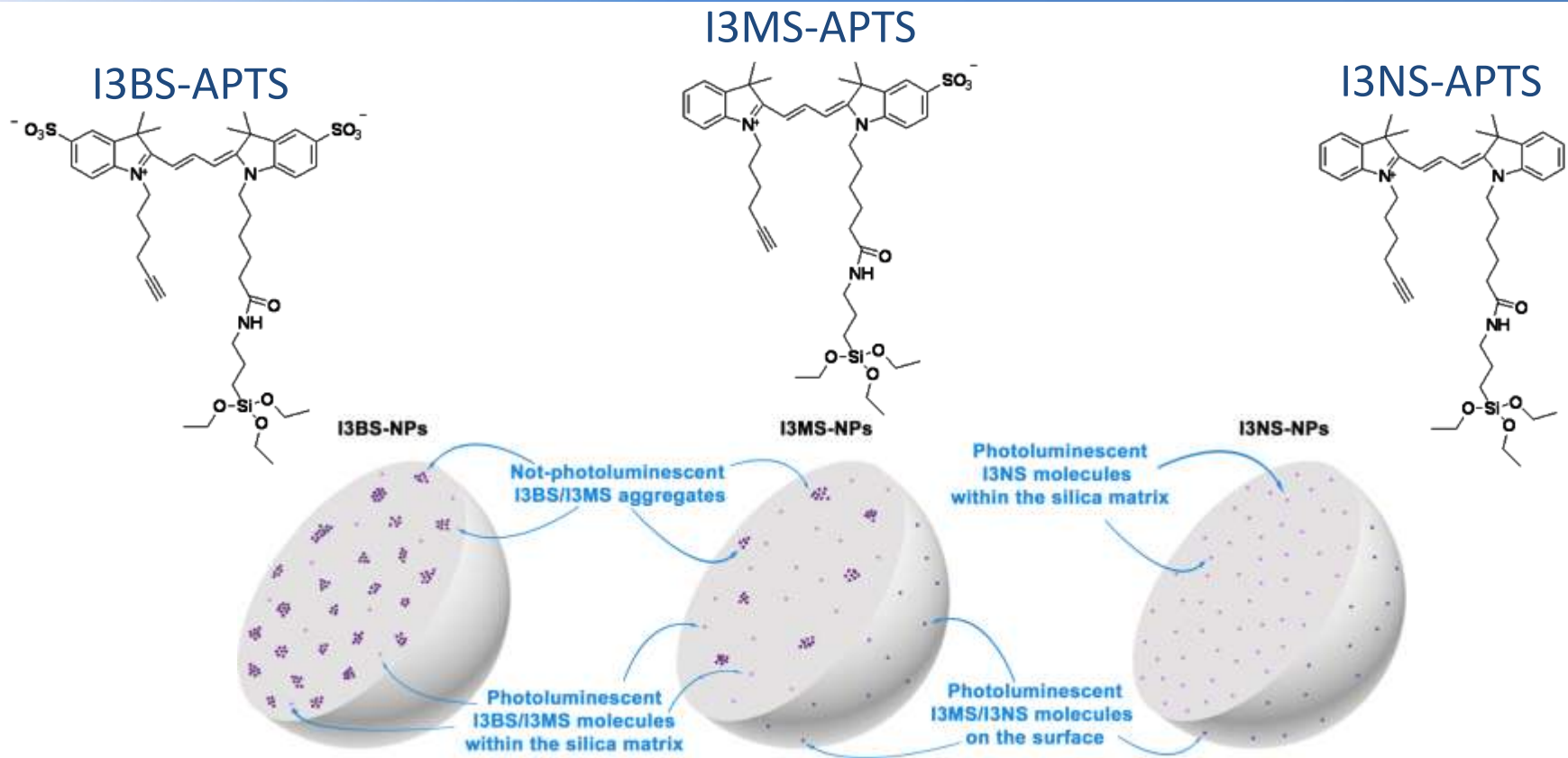
% emitting Cy

0.5

63

97

Photoluminescence performances of dyes in NPs vs dye in solution



Cyanines per NP		
110	86	110
Average Photoemission Gain per Cy		
0.04	2.7	4.3
Overall Brightness Enhancement per NP		
4.4	232	481

Conclusions

The control of molecular parameters can actually result in a virtuous design of the photophysical behaviour of highly homogeneous hybrid dye-silica NP: “wise” optimization of the use of organic fluorophores

The interplay between steady state and time resolved photoluminescence actually represents an unique tool to monitor and understand the “molecular story” leading to the final materials

For further details you can refer to G.Alberto et al. *Chem. Mater.* 2012, 24, 2792-2801

Perspectives

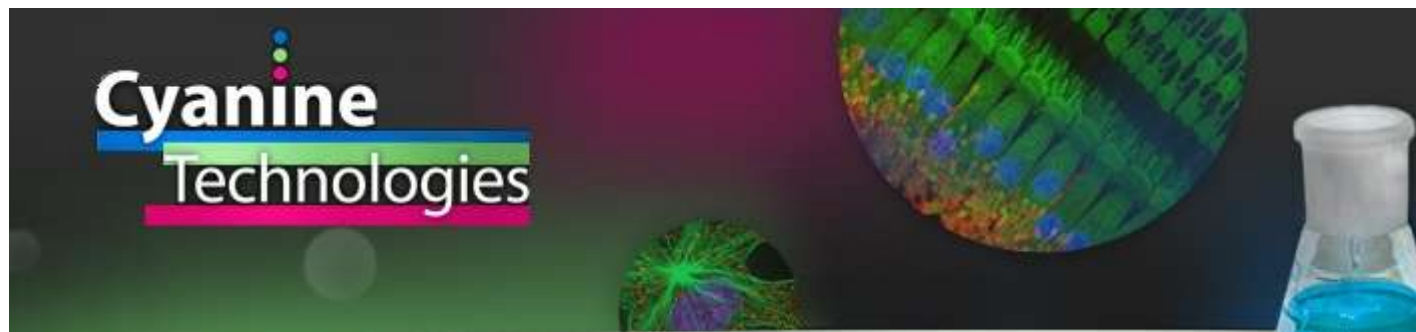
Possibility to increase the number of dye molecules per NP keeping the optimized dye distribution, location and photoluminescence intensity

Possibility to consider other photoluminescent molecules with higher Φ

Possibility to consider other photoluminescent molecules emitting in the Red-NIR boundary region



www.nis.unito.it

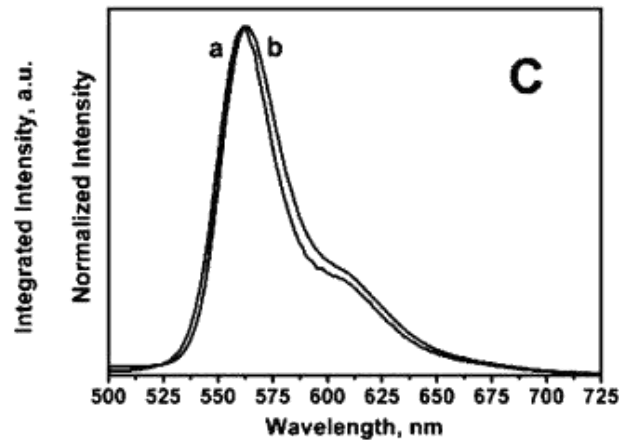
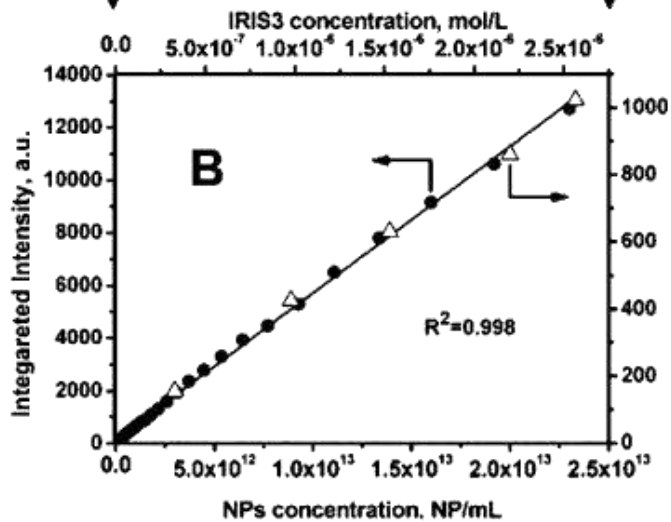
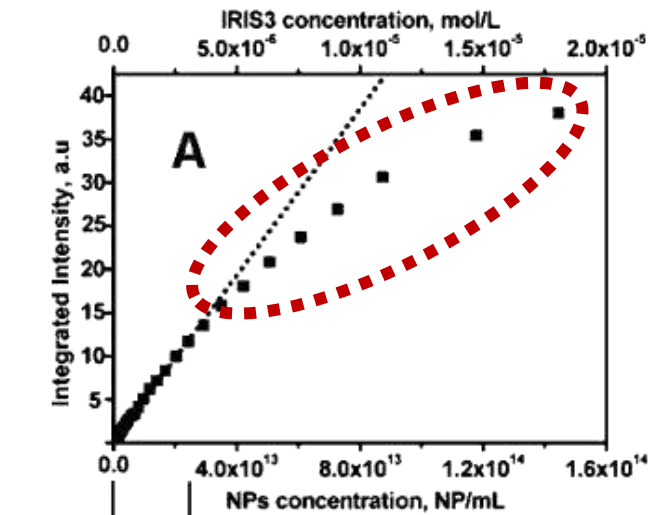


Photophysical performances of hybrid dyes/silica NP

Intensity: pay attention to interparticles effects!

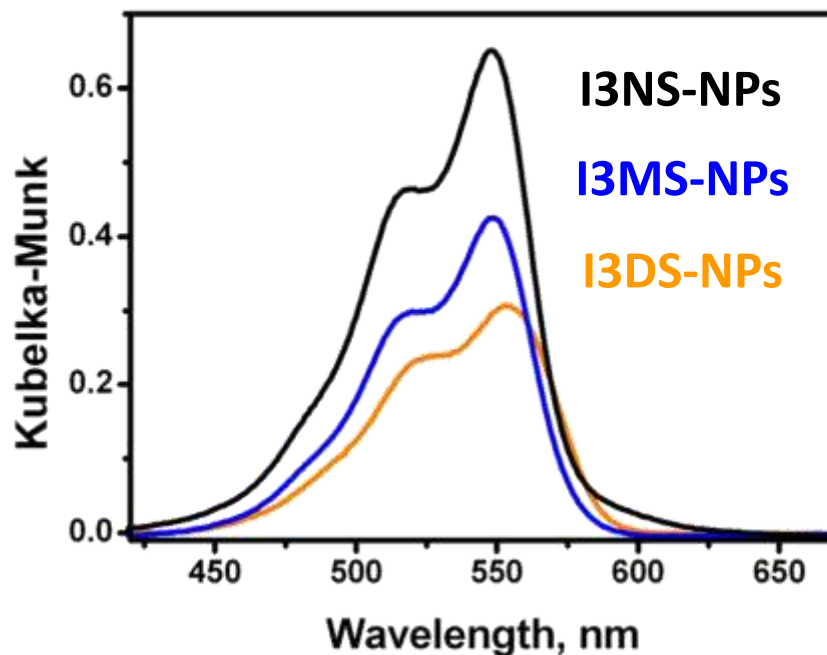
from diluted to concentrated suspensions

- light scattering
- inner filter effects
- self-absorption



Where are the missing, not photoluminescent cyanines ?

Trace: space



I3DS-NPs



I3MS-NPs



I3NS-NPs



prevalence of diffusion towards adsorption