

Nanocomposites and nanomaterials

CRYSTAL STRUCTURE OF THE TbCu₅Al₇ COMPOUND

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Rare-earth based materials are one of the most investigated due to their widespread potential applications such as permanent magnets [1], magnetocaloric systems [2], hydrogen storage materials [3] and other applications. Crystallite size reducing up to nano-scale provides an opportunity to improve the properties of material [4, 5]. Therefore, investigation of new phases as a first step of the nano material preparation is important task for researcher. In this work we present results of investigations of the TbCu₅Al₇ ternary compound.

In references [6] the authors presented information concerning the existence of the TbCu₅Al₇ ternary compounds and the ThMn₁₂ structure type was proposed without any crystallographic data.

The sample was prepared by arc melting of the elements with further annealing at 600°C. Crystal structure of the this compound was successfully refined by the Rietveld method using the X-ray powder pattern of the TbCu₅Al₇ sample collected on the Guinier Huber G 670 diffractometer (CuK α ₁ radiation). The structure was successfully solved in the $I4/mmm$, ThMn₁₂ structure type, Pearson symbol, $tI26$, $a = 8.70178(4)$, $c = 5.12573(3)$ Å; $R_1 = 0.0467$; $R_p = 0.0780$; $R_{wp} = 0.0818$. Atomic coordinates as well as isotropic displacement parameters are summarized in Table.

Tabl. 1. Atomic coordinates and isotropic displacements parameters of the TbCu₅Al₇ compound

Atom	Site	x	y	z	$B_{iso}, \text{\AA}^2$
Cu1	8f	1/4	1/4	1/4	0.66(2)
M^*	8j	0.2803(1)	1/2	0	0.46(4)
Al	8i	0.3451(2)	0	0	0.56(4)
Tb	2a	0	0	0	0.54(2)

*Mixed occupation.

$$M = 0.143(3) \text{ Cu} + 0.857(3) \text{ Al}$$

Crystal structure of this compound can be consider as “cluster” formed from Cu and Al atoms around Tb. Packing 20 vertex polyhedrons are represented in Figure.

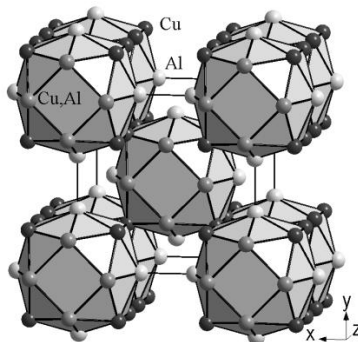


Fig. Polyhedron packing in the structure of TbCu₅Al₇ ternary compound

1. Gutfleisch O. et al. Magnetic Materials and Devices for the 21st Century: Stronger, Lighter, and More Energy Efficient Adv.Mater. 23 (2011) 821.
2. Pecharsky V. K. and Gschneidner K. A. Giant Magnetocaloric Effect in Gd₅(Si₂Ge₂) Phys. Rev.Lett. 78 (1997) 4494.
3. Jena P. Materials for Hydrogen Storage: Past, Present, and Future J.Phys.Chem.Lett. 2 (2011) 206.
4. Berlouis L.E. et al. Thermal analysis investigation of hydriding properties of nanocrystalline Mg-Ni- and Mg-Fe-based alloys prepared by high-energy ball milling J Matter. Res., 16 (1) (2011) 45
5. Akdogan O. et al. Effect of Exchange Interactions on the Coercivity of SmCo₅ Nanoparticles Made by Cluster Beam Deposition Adv. Funct. Mat. 23 (2013) 3262
6. Felner I. Crystal structures of ternary rare-earth-3d transition metal compounds of the RT₆Al₆ type. J. Less-Comm. Met. 72 (1980) 241