

Catalytic effect of nanostructure SiC and Ni on Mg-based hydrogen storage materials prepared by reactive mechanical alloying method

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Among the metal hydrides, magnesium has the theoretically highest weight capacity for hydrogen storage (7.6 wt.%), lightweight and a reasonably low cost [1]. However, high working temperature, slow reaction kinetics and difficult activation limit the practical application of Mg-based hydrides. Many efforts have been done to improve sorption properties and reaction kinetics such as element substitution (metal or metal oxides) as catalyst in nanometer scale and modification of ball milling technique as well [2-4]. Recently, the reactive ball milling under hydrogen atmosphere was also successfully introduced to prepare hydrogen storage materials [5].

In this work we success to synthesis and investigate the catalytic effect of SiC and Ni (in nanostructure scale) on MgH₂ using reactive mechanical alloying method in 10 bar H₂. At first step, using SiC catalyst the sorption properties can be improved. The most promising step by using nano-catalysts of SiC and Ni (MgH₂-5wt%SiC-5wt%Ni) which absorb 5.7 wt% hydrogen and at the same time decrease the desorption temperature to 250°C. Compared to T onset of pure MgH₂ -which desorp at 380°C.

Keyword: Magnesium, hydrogen storage, reactive mechanical alloying, catalyst, silicon carbide.

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