

- **Nanocomposites and nanomaterials**

## Synthesis and characterization of $(\text{Cr}_x\text{V}_{1-x})_{n+1}\text{AlC}_n$ solid solution

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MAX phases are a class of ternary nitrides and carbides, with the general formula  $\text{M}_{n+1}\text{AX}_n$ . These phases have a hexagonal crystal structure that consists of edge-sharing  $[\text{M}_6\text{X}]$  octahedron interleaved with A layers; they are thus considered as nanolaminated-layered materials with a unique combination of metal and ceramic properties [1]. For example, they are resistant to oxidation and corrosion, elastically stiff, but at the same time, they also demonstrate high thermal and electrical conductivities and they are machinable.

$(\text{Cr}_x\text{V}_{1-x})_{n+1}\text{AlC}_n$  MAX phases solid solutions with  $n=1, 2, 3$  and  $x=0, 0.25, 0.5, 0.75, 1$  were obtained from commercial Cr, V, Al and C powders by using pressureless reactive sintering at high temperature (1400°C to 1600°C) or hot isostatic pressure technique (80MPa, 1450°C).

X-ray powder diffraction analysis confirmed the existence of continuous  $(\text{Cr}_x\text{V}_{1-x})_2\text{AlC}$  solid solution, the amounts of additional phases increasing with the synthesis temperature and the V content. The synthesis of  $(\text{Cr}_x\text{V}_{1-x})_3\text{AlC}_2$  and  $(\text{Cr}_x\text{V}_{1-x})_4\text{AlC}_3$  phases was realized only for  $x=0.25; 0.5; 0.75$  and  $x=0; 0.25; 0.5$ , respectively.

Structural parameters of all modifications of  $(\text{Cr}_x\text{V}_{1-x})_{n+1}\text{AlC}_n$  solid solution were established by full profile Rietveld refinements. Lattice parameters and cell volumes decrease linearly with increasing the Cr content, while the values of z positions (free parameters in the unit cell) remain almost constant. The evolutions of the interatomic and interlayer distances as a function of x and n have been constructed.

1. Barsoum M. W. The  $\text{M}_{n+1}\text{AX}_n$  phases: A new class of solids thermodynamically stable nanolaminates // Progress in Solid State Chemistry.-2000.- **28**.-P. 201-281.