

## Nanocomposites and nanomaterials

### Nanosized Pt/SnO<sub>2</sub> materials for adsorption semiconductor sensors for detection CO in air

I.P. Matushko, N.P. Maksymovych, L.P. Oleksenko, H.O. Arinarkhova

*Faculty of Chemistry, Taras Shevchenko National University of Kyiv  
12 Lva Tolstogo Str., Kyiv 01033, Ukraine  
E-mail: Anna.arinarkhova@gmail.com*

Nowadays increasing pollution of carbon monoxide in air due to transport and industrial emissions is observed. Since CO is a toxic gas, it is necessary to create different devices for detection of carbon monoxide in air. Adsorption semiconductor sensors because of their properties are very useful to use them as a detector of such devices. Sensitivity of the sensor toward carbon monoxide may be enhanced by adding catalysts to increase a rate of an oxidation reaction of CO by oxygen on the surface of a sensitive layer of the sensor. That is why the aim of the work was to obtain Pt/SnO<sub>2</sub> nanomaterials and study their gas sensitive properties to CO.

Nanosized material SnO<sub>2</sub> was prepared by a sol–gel method using SnCl<sub>4</sub>•5H<sub>2</sub>O and ethylene glycol. It was shown by TEM method that particle size of SnO<sub>2</sub> nanomaterial ranged from 5 to 30 nm (an average particle size was 17 nm). Only reflexes of the cassiterite phase were found in the analyzed nanomaterial using XRD method. Sensors with different Pt contents were created on the base of the nanosized tin dioxide nanomaterial by its impregnation with H<sub>2</sub>PtCl<sub>6</sub> solutions of different concentrations ( $0.21 \times 10^{-2}$  -  $35 \times 10^{-2}$  M). It was found that electrical resistances of the sensors increase with increasing the platinum content in the sensitive material of the sensors. A ratio of the electrical resistance of the sensor in air ( $R_0$ ) to its electrical resistance in the analyzed gas with concentration of 900 ppm CO in air ( $R_g$ ) was chosen as a measure of the sensor response ( $\gamma = R_0/R_g$ ).

It was found that introducing platinum additives to the sensor sensitive materials increases the responses of the sensors in a great extent. An extremal dependence of the sensor response on concentration of H<sub>2</sub>PtCl<sub>6</sub> solution was found. Besides, it was observed that maximums of the sensor responses were shifted to the region of lower concentrations of the H<sub>2</sub>PtCl<sub>6</sub> solutions when temperatures of the sensors were increased. The highest sensors response to CO ( $\gamma = 29.1$ ) is achieved for the sensor obtained by impregnation of nanosized SnO<sub>2</sub> with a solution of  $0.84 \times 10^{-2}$  M H<sub>2</sub>PtCl<sub>6</sub> at the sensor temperature corresponding to its heater power consumption of 0.25 W.

A study of the catalytic activities of the sensor nanomaterials showed that their activities increased with increasing quantity of Pt in the content of the materials. The created sensors possess the fast response and recovery time.