

Nanocomposites and nanomaterials

Adsorption of U(IV) by mono- and bifunctional mesoporous silica with phosphonic and carboxylic groups

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Mono- and bifunctional mesoporous silica with phosphonic ($-\text{PO}(\text{OH})_2$) and carboxylic ($-\text{COOH}$) groups in the surface layer were synthesized via one-pot template method. The samples are characterized by well-developed surface (above $500 \text{ m}^2/\text{g}$), and the concentration of functional groups about $1\text{--}2 \text{ mmol/g}$.

The new approach to the synthesis of phosphorus-containing silica increased the U(IV) sorption capacity (by 1.5 times) and distribution factor (in order of magnitude) compared to samples described in [1], with respective values of 260 mg/g and $3,3 \cdot 10^4 \text{ cm}^3/\text{g}$ (at pH 2).

Simultaneous binding of carboxylic and phosphonic groups does not deteriorate the ordered structure of the synthesized materials, but the values of distribution factor and sorption capacity decrease compared with mono-functional samples, which may be due to the adverse reaction of carboxylic groups with the template (Pluronic P123).

The obtained samples were analyzed by a number of physical and chemical methods (nitrogen adsorption, X-ray, IR and NMR spectroscopy, TEM, SEM).

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1. *V.V. Milyutin, V.M. Gelis, N.A. Nekrasova et al. Sorption of Actinide Ions onto Mesoporous Phosphorus-Containing Silicas // Radiochemistry. –2014. –56. –P. 262–266.*