

Physico-Chemical nanomaterials science

Electrokinetic properties of colloidal silica nanoparticles with magnetic core

K. Gdula¹, M. Barczak¹, E. Skwarek²

¹*Department of Theoretical Chemistry, Faculty of Chemistry,
Maria Curie-Skłodowska University, Maria Curie-Skłodowska Sq. 3, 20 031
Lublin, Poland.*

E-mail: karolina.gdula@poczta.umcs.lublin.pl

²*Department of Radiochemistry and Colloid Chemistry, Faculty of Chemistry,
Maria Curie-Skłodowska University, Maria Curie-Skłodowska Sq. 3, 20 031
Lublin, Poland.*

Many works are devoted to electrokinetic properties of α -Fe₂O₃ oxides [1]. In this work we present properties of other iron oxides, such as: γ -Fe₂O₃ or Fe₃O₄ and their nanocomposites. Three-step reaction was carried out in order to obtain functionalized silica colloids with magnetic properties. Aqueous suspension of iron oxides nanoparticles as magnetic core (mainly maghemite – γ -Fe₂O₃) were obtained by Massart's method [2]. A silica shell was obtained onto maghemite employing a modified sol-gel method by Salgueirino [3].

In this work we present electrokinetic properties of the synthesized nanoparticles composed of magnetic core and silica shell. Effect of pH and ionic strength on the electrokinetic properties of the above-mentioned nanocomposites were examined. The obtained results show that with increasing pH, value of zeta potential (of all tested nanocomposites) decreases. All measurements were carried out using dynamic light scattering technique, in a ZetaSizer 3000 (Malvern, U.K.).

Acknowledgment

The research leading to these results has received funding from the People Programme (Marie Curie Actions) of the European Union's Seventh Framework Programme FP7/2007-2013/ under REA grant agreement n° PIRSES-GA-2013-612484.

1. Janusz W., et.al. Electrical double layer at the α -Fe₂O₃-mixed electrolyte (ethanol-aqueous) interface // J Colloids Surf A.-1999.-**149**.-P.421-426.
2. Massart R. Preparation of Aqueous Magnetic Liquids in Alkaline and Acidic Media // IEEE Trans Magn.-1981.-**MAG-17**.-P.1247-1248.
3. Salgueirino Maceira V., et.al. Bifunctional gold-coated magnetic silica

spheres // Chem Mater.-2006.-**18**.-P.2701-2706.