

# Analysis of the effect of $\text{AlCl}_3$ and $\text{FeCl}_3$ concentration in BSA solutions on the parameters of film textures upon reaching the isoelectric point

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## Introduction

Previously, we have presented a method for assessing the effect of physicochemical factors on biopolymers by analyzing textures on films resulting from their dried solutions [1]. The main parameters for describing textures in this method are the specific length  $L_{\text{spec}}$  and the number of segments  $N_{\text{segm}}$  of zigzag patterns (Fig. 1).

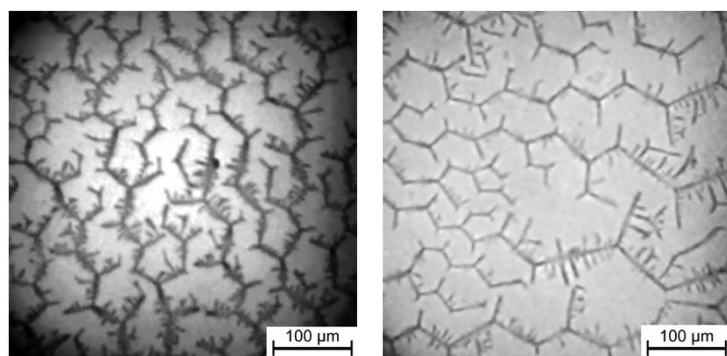


Fig. 1. Example micrographs for films obtained from 0.5 mg/ml BSA + 20 mmol/l NaCl solutions.

## Methods

Sample solutions were prepared in distilled water, using 20 mmol/l NaCl, 0.5 mg/ml BSA (DiaM, USA), and a series of concentrations (from 0.1 to 0.4 mmol/l) of  $\text{AlCl}_3$  or  $\text{FeCl}_3$ . The films were obtained by drying 0.5 ml of each sample in 20×20 mm<sup>2</sup> glass cells, in a low-pressure thermostatic environment. pH measurements were carried out using the pH-150MI pH meter (Izmeritelnaya Tekhnika, Moscow).

## Results

Analysis of the influence of different concentrations of  $\text{AlCl}_3$  and  $\text{FeCl}_3$  on the change in texture parameters has shown that in the pH range from 3.58 to 3.38, the decrease in  $N_{\text{segm}}$  and  $L_{\text{spec}}$  for  $\text{AlCl}_3$  ranges from 5000 to 1800 and from 4.2 to 2.6, respectively. Fig. 2 demonstrates typical textures for the films with 0.3 mmol/l  $\text{AlCl}_3$ .

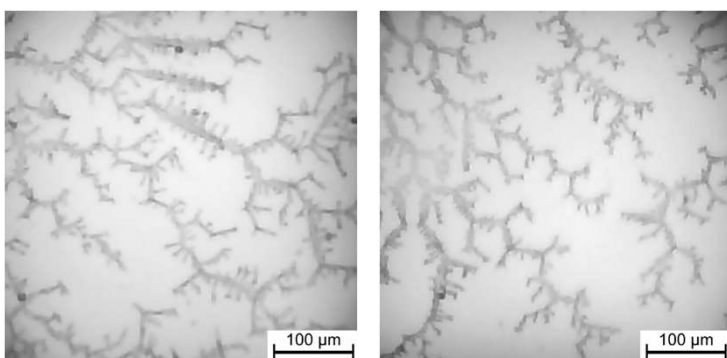


Fig. 2. Example micrographs for films obtained from 0.5 mg/ml BSA + 20 mmol/l NaCl + 0.3 mmol/l  $\text{AlCl}_3$  solutions (pH ≈ 3.43).

In the pH range from 2.8 to 2.46, the decrease in  $N_{\text{segm}}$  and  $L_{\text{spec}}$  for  $\text{FeCl}_3$  is from 6400 to 0 and from 6 to 0, respectively. Fig. 3 demonstrates typical textures for the films with 0.3 mmol/l  $\text{FeCl}_3$ .

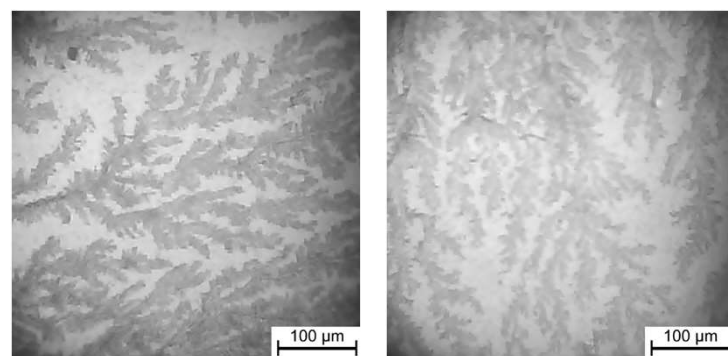


Fig. 3. Example micrographs for films obtained from 0.5 mg/ml BSA + 20 mmol/l NaCl + 0.3 mmol/l  $\text{FeCl}_3$  solutions (pH ≈ 2.54).

The decrease in  $N_{\text{segm}}$  and  $L_{\text{spec}}$  correlates with the onset of BSA precipitation in solutions at pH < 3.7 ( $\text{AlCl}_3$ ) and pH < 3 ( $\text{FeCl}_3$ ), which may be due to the reentrant condensation in solutions due to a significant decrease in the absolute value of the protein surface potential [2].

## Conclusion

Thus, it can be concluded that the dominant factor in the change in  $N_{\text{segm}}$  and  $L_{\text{spec}}$  is the change in the concentration of  $\text{Al}^{+3}$  and  $\text{Fe}^{+3}$  ions.

## References

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