

# Positronics in contemporary nanocomposites science and engineering: the case of multinanoparticulate substances

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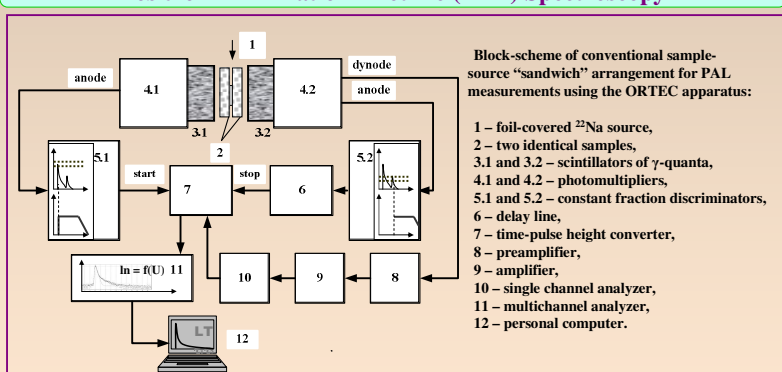
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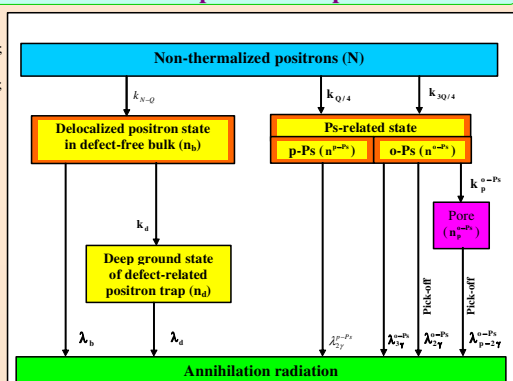
## EXPERIMENTAL:

### Positron Annihilation Lifetime (PAL) Spectroscopy

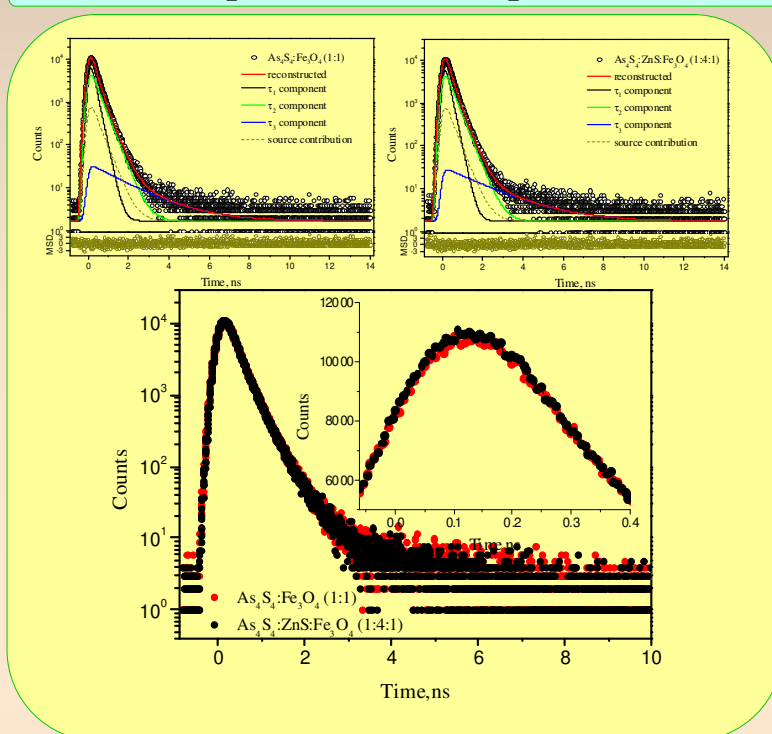


### Mixed channels of positron trapping and o-Ps decaying in unconstrained x3-term decomposed PAL spectra

$n^{p-Ps}$  – number of p-Ps at time  $t$ ;  
 $\lambda^{p-Ps}$  –  $2\gamma$ -quanta annihilation rate of p-Ps;  
 $n^{o-Ps}$  – number of o-Ps at time  $t$ ;  
 $\lambda^{o-Ps}$  –  $3\gamma$ -quanta annihilation rate of o-Ps;  
 $\lambda^{p-o-Ps}$  – annihilation rate of pick-off o-Ps;  
 $n^{p-o-Ps}$  – number of pores within o-Ps is formed;  
 $\lambda^{p-o-Ps}$  – o-Ps annihilation rate in pick-off process through pores;  
 $\lambda^{p-o-Ps}$  – positron trapping rate of pore.  
 $\tau_{av} = \frac{\tau_1 I_1 + \tau_2 I_2 + \tau_3 I_3}{I_1 + I_2 + I_3}$   
 $\tau_3 = [1 - \frac{R}{R + \Delta R} + \frac{1}{2\pi} \sin(\frac{2\pi R}{R + \Delta R})]^{-1}$   
 $V_f = \frac{4\pi R^3}{3}$   
 $R$  – pore radius;  
 $\Delta R = 0.166 \text{ nm}$  – fitted empirical electron layer thickness;  $V_f$  – free volume size.



## Experimental PAL spectra



PAL spectra of multinanoparticulate arsenical-based systems reconstructed from unconstrained x3-fitting: biparticulate  $As_4S_4/Fe_3O_4$  and triparticulate  $As_4S_4/ZnS/Fe_3O_4$ .

## Conclusions:

Possibilities of positron annihilation lifetime (PAL) spectroscopy applied to characterize nanosization processes under high-energy mechanical milling are analyzed for multiparticulate arsenical-based systems, such as monoparticulate  $As_4S_4$ , biparticulate  $As_4S_4/Fe_3O_4$  and triparticulate  $As_4S_4/ZnS/Fe_3O_4$ . The algorithm to treat registered PAL spectra for such substances within three-state mixed trapping model evolving competitive channels of positron and positronium Ps (bound electron-positron state) channels (x3-x2-CDA, coupling decomposition algorithm) is given. It is shown that coexistence of nanocrystalline  $As_4S_4$  phase and supplemented amorphous substance is crucial feature of these materials, the latter being generated owing to *reamorphization* of disordered phase initially existed in arsenic sulphide prepared by conventional synthesis from elemental precursors and direct milling-driven vitrification of nanocrystalline  $As_4S_4$  phase.

## PAL spectra fitting within unconstrained x3-decomposition

Sample	fit-11	Fitting parameters										Component input					Trapping modes			Volume $R_3$ nm	$f_3$ %
		$\tau_1$ ns	$I_1$ a.u.	$\tau_2$ ns	$I_2$ a.u.	$\tau_3$ ns	$I_3$ a.u.	$\tau_{av}^1$ ns	$\tau_{av}^2$ ns	$\tau_{av}^3$ ns	$\tau_{av}^4$ ns	$\tau_{av}^5$ ns	$\tau_{av}^6$ ns	$\tau_{av}^7$ ns	$\tau_{av}^8$ ns	$\tau_{av}^9$ ns	$\tau_{av}^{10}$ ns	$\tau_{av}^{11}$ ns			
$As_4S_4/Fe_3O_4$ 1:1	0.07	0.215	0.695	0.406	0.297	1.995	0.008	0.149	0.121	0.017	0.287	0.272	0.250	0.66	0.16	1.62	0.288	0.15			
	0.07	0.222	0.732	0.419	0.260	2.278	0.007	0.162	0.109	0.017	0.288	0.273	0.253	0.56	0.17	1.66	0.313	0.17			
$As_4S_4/ZnS$ 1:4	0.02	0.239	0.749	0.434	0.234	2.145	0.016	0.179	0.102	0.035	0.316	0.285	0.267	0.45	0.17	1.62	0.301	0.34			
	0.03	0.238	0.753	0.440	0.231	2.207	0.015	0.179	0.102	0.034	0.315	0.285	0.267	0.45	0.17	1.65	0.307	0.34			
$As_4S_4/ZnS/Fe_3O_4$ 1:4:1	0.01	0.219	0.689	0.410	0.302	2.242	0.009	0.151	0.124	0.020	0.294	0.277	0.255	0.65	0.15	1.61	0.309	0.20			
	0.03	0.221	0.709	0.419	0.282	2.610	0.008	0.157	0.118	0.021	0.296	0.278	0.256	0.61	0.16	1.64	0.339	0.24			
$As_4S_4/Fe_3O_4$ 1:1	0.05	0.191	0.486	0.374	0.506	2.071	0.009	0.093	0.189	0.018	0.300	0.284	0.254	1.31	0.12	1.47	0.294	0.17			
	0.06	0.193	0.485	0.372	0.507	2.191	0.008	0.094	0.188	0.018	0.301	0.284	0.256	1.27	0.12	1.45	0.305	0.18			
$As_4S_4/ZnS/Fe_3O_4$ 1:4:1	0.01	0.191	0.454	0.370	0.536	1.929	0.010	0.087	0.198	0.020	0.305	0.288	0.259	1.37	0.11	1.43	0.281	0.17			
	0.01	0.193	0.463	0.372	0.527	1.948	0.010	0.089	0.196	0.020	0.305	0.288	0.259	1.33	0.11	1.43	0.283	0.17			
	0.01	0.200	0.498	0.380	0.492	2.027	0.010	0.100	0.187	0.020	0.306	0.289	0.261	1.17	0.12	1.45	0.291	0.18			
0.02	0.200	0.504	0.381	0.488	2.285	0.009	0.101	0.186	0.020	0.307	0.289	0.261	1.17	0.12	1.46	0.314	0.21				
$As_4S_4/ZnS$ 1:4	0.04	0.190	0.360	0.353	0.620	1.816	0.020	0.068	0.219	0.036	0.324	0.293	0.268	1.54	0.08	1.32	0.271	0.30			
	0.04	0.194	0.378	0.356	0.602	1.835	0.020	0.073	0.214	0.036	0.324	0.294	0.269	1.44	0.09	1.32	0.272	0.30			
	0.04	0.200	0.408	0.361	0.573	1.867	0.019	0.082	0.207	0.036	0.324	0.294	0.271	1.30	0.09	1.34	0.276	0.30			
0.05	0.200	0.428	0.369	0.555	2.234	0.017	0.086	0.205	0.038	0.329	0.295	0.270	1.29	0.10	1.37	0.308	0.38				

## PAL spectra fitting within x3-x2-CDA

Effect of ZnS: AZF-141 =  $As_4S_4 : ZnS : Fe_3O_4$  (1:4:1) irt AF-11 =  $As_4S_4 : Fe_3O_4$  (1:1)  
 Effect of  $Fe_3O_4$ : AZF-141 =  $As_4S_4 : ZnS : Fe_3O_4$  (1:4:1) irt AZ-14 =  $As_4S_4 : ZnS$  (1:4)

Samples	$\tau_n$ [ns]	$I_n$ [a.u.]	$\tau_{int}$ [ns]	$I_{int}$ [a.u.]	$\tau_{av1}$ [ns]	$\tau_{av2}$ [ns]	$\tau_{av3}$ [ns]	$\tau_n$ [ns]	$\tau_{av}$ [ns]	$\tau_2/\tau_1$	$\tau_3/\tau_1$
Effect of ZnS:											
AZF-141 irt AF-11	0.190	-0.027	0.456	-0.026	-0.005	-0.012	0.320	0.266	1.500±1.50	0.190	1.713±1.7
AF-11 irt AZF-141	0.190	0.028	0.456	0.024	0.005	0.011	0.312	0.260	1.412±1.41	0.196	1.754±1.7
Effect of $Fe_3O_4$ :											
AZF-141 irt AZ-14	0.189	0.265	0.364	0.316	0.050	0.115	0.285	0.256	1.380±1.38	0.108	1.421±1.4

