НАН УКРАІНИ EPR study of nanosized fumed silica $I\Phi H$ incorporated with aged Zn(acac), ethanol NAS UKRAINI Fyzikální ústav Akademie věd České republiky FZU solution

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Introduction

Typically, pure pyrocarbon in hybrid carbonized silica (C/SiO₂) adsorbents possesses a relatively small number of active sites (oxidized groups) which could play an essential role in the bonding of polar molecules, as the carbon layer has mainly a non-polar pregraphite structure with the size of basal planes in graphene particles of several nanometers. To increase the amounts of active sites on hybrid adsorbent surfaces, additional oxidizing of pyrocarbon with the formation of COH, C=O, COOH, etc. groups, mixed X/SiO2 oxides as substrates (possessing a larger number of active sites than parent silica has), or organometallics as precursors can be utilized. The presence of carbon-metal compounds can change not only the topology but also other characteristics of C/X/SiO2 adsorbents, such as surface site distribution and catalytic and adsorptive abilities. From the side, the functionalization of silica with aged $Zn(acac)_2$ solution that possesses intensive photoluminescence in the visible range can be perspective for the fabrication of modern luminophores. Therefore the study of the electronic structure of the defects is crucial for potential applications of this novel nanocomposites.

Materials and methods

The fumed silica (S~295 m²/g, d~10-12 nm) was infiltrated with aged luminescent Zn(acac)₂ ethanol solution of different concentrations (1 wt.% and 4 wt.%). As a result, SiO₂:C nanocomposites with zinc silicate (Zn₂SiO₄) layer were obtained. The samples were annealed in the air at 600°C. The electron paramagnetic resonance (EPR) spectra were measured at X-band frequency range (v~9.4 GHz) on Bruker ELEXSYS E580 spectrometer.

Zinc acetylacetonate C₁₀H₁₄O₄Zn

Nanosized fumed silica

Carbon-mineral adsorbent: CS_{7n} ISP



EPR spectra decomposition

The single EPR line at g=2.0040(3) dominates the EPR spectra, and its shape varies from Lorentzian in the initial CS_{Zn} nanocomposites prepared with 4% of Zn(acac)₂ to Gaussian after thermal annealing at 600°C. The EPR signal at g=2.0025(3) with temperaturedependent linewidth and intensity was detected as well. In addition, a weak broad signal at g=2.0051(3) was observed.

4% Zn(acac)₂, T_{ann}=600°C



Spin concentration of the paramagnetic centers in CS_{Zn} nanocomposites at T=295 K

In the CS_{Zn} nanocomposites prepared with 1% of $Zn(acac)_2$, no EPR spectra were observed. After thermal annealing of those samples at 600°C, the spin concentration of the paramagnetic centers at 295 K was very low: $N_{s} \sim 5.4 \cdot 10^{9}$ spins/mm³.

In the CS_{7n} nanocomposites prepared with 4 wt.% of $Zn(acac)_2$, the paramagnetic centers had $N_{s} \sim 3.1 \cdot 10^{11}$ spins/mm³, while the thermal annealing at 600°C doubles it: $N_{S} \sim 6.7 \cdot 10^{11}$ spins/mm³.

EPR spectra assignment

The weak EPR signal with g=2.0051(3) and Lorentzian lineshape can be tentatively attributed to silicon dangling bonds that can be formed at the interface of Zn₂SiO₄ layer and carbonized silica or at the Zn₂SiO₄ surface layer.

The intense signal at g=2.0040(3) should be related to carbon-related defect with an adjacent oxygen atom. The transformation of its EPR lineshape from Lorentzian to Gaussian one with the increase of the thermal annealing temperature can be explained by the appearance of the superhyperfine interaction of this defect with surrounding nuclei (e.g. ¹H).



The third Lorentzian EPR signal with g=2.0025(3) can be assigned to carbon-dangling bonds in the sp^2 hybridized state. At the same time, no EPR signals that can be related to the defects usually observed in ZnO particles were detected.

Conclusions

EPR spectra consist of three paramagnetic centers: two of them are related to carbon-related centers located in uniform carbon layer that cover silica nanoparticles, and another one is attributed to silicon dangling bonds raised from the formation of Zn_2SiO_4 layer.

The spin concentration of the paramagnetic centers rises with the increase of $Zn(acac)_2$ concentration and annealing temperature.

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