

Application of the Rutherford ion backscattering spectrometry method in powder nanotechnology: YSZ – system.

Doroshkevich A.S.^{1,3}, Tatarinova A.A.^{1,2}, Kulik M.^{1,4}, Shylo A.V.³,
Zuk J.⁴, Budzyński M.⁴, Lyubchik A. I.⁵, Cornei N.⁶, Mardare D.M.⁶, Mita C.⁶, Chicea D.⁷

¹ Joint Institute for Nuclear Research, 141980, str. Joliot-Curie, 6, Dubna, Russia

² Dubna State University, 141980, st. Universitetskay, 19, Dubna, Russia

³ Donetsk Institute for Physics and Engineering named after O O Galkin NAS of Ukraine, Nauki ave, 46, Kyiv, Ukraine

⁴ Institute of Physics, Maria Curie-Skłodowska University, Pl. Marii Curie-Skłodowskiej 1, 20-031 Lublin, Poland

⁵ i3N/CENIMAT, Department of Materials Science, Faculty of Science and Technology, New University of Lisbon and CEMOP/UNINOVA, Campus de Caparica, Caparica, Portugal

⁶ Alexandru Ioan Cuza University of Iasi, Faculty of Physics, 700506, Bld. Carol I, No. 11, Iasi, Romania

⁷ University "LUCIAN BLAGA" of SIBIU (ULBS) 550012, Str. Dr. Ion Ratiu 7-9 Sibiu, Romania

Introduction

Rutherford Backscattering Spectrometry (RBS) is an ion scattering technique used for compositional thin film that are less than 1 μ m thick analysis. During an RBS analysis, high-energy He²⁺ ions with energies in the region from several hundred kiloelectron-volts to 2 - 3 MeV are directed onto the sample and the energy distribution and yield of the backscattered He²⁺ ions at a given angle is measured. Since the backscattering cross section for each element is known it is possible to obtain a quantitative compositional depth profile.

The capabilities of this method can be significantly expanded. In particular, the method can be used in powder nanotechnology to study elemental composition in microscopically small objects.

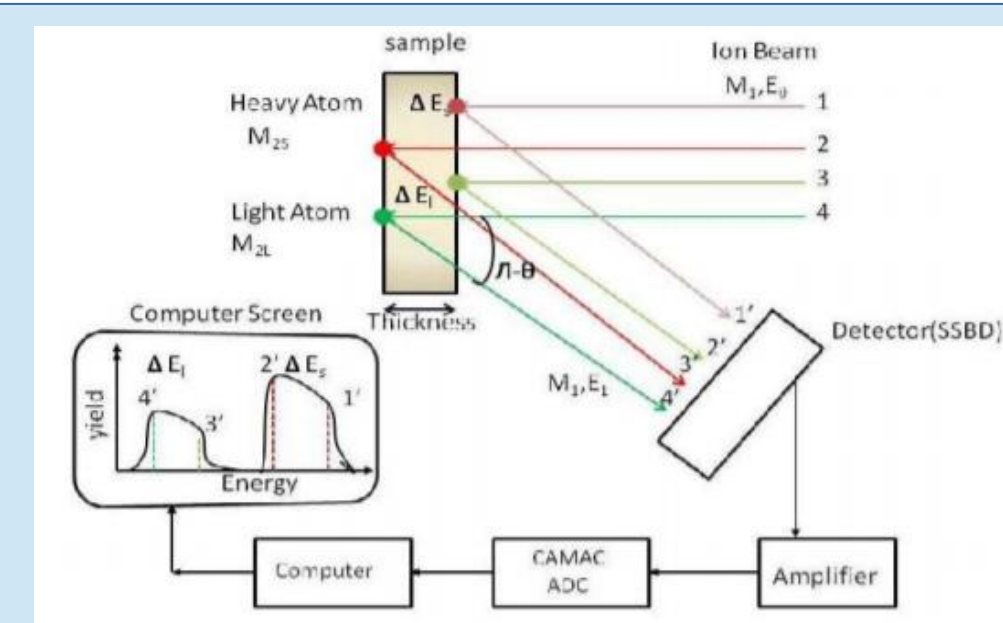


Figure.1. Schematic diagram of RBS facility [2].

Methods and Materials

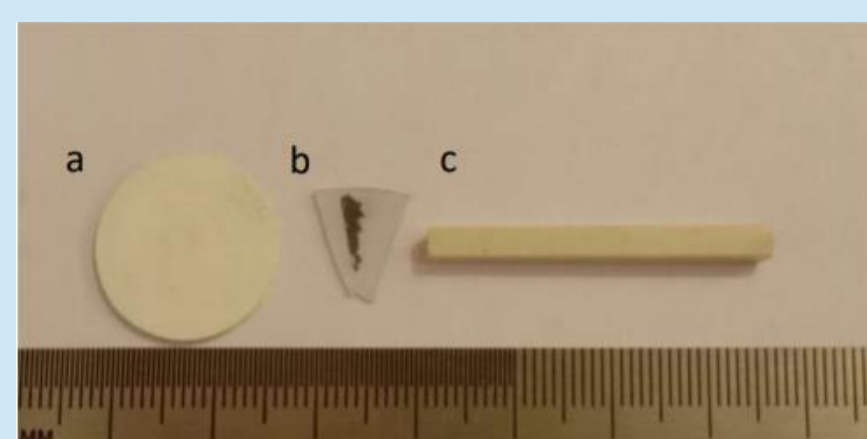


Figure 2. Samples of investigation: a - functional environment for volumetric chemo-electronic converter – tablet; b - Single crystal ZrO₂ with juvenile surface; c - Polycrystalline ZrO₂ with polished surface.

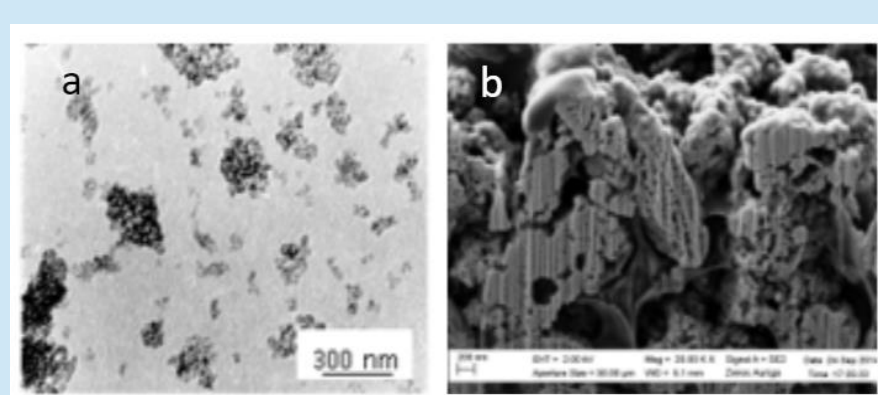


Figure 3. a - Transmission electron microscopy (TEM) images of nano-powder composition ZrO₂-3mol%Y₂O₃, 400°C. 2Hours; b - Image scanning electron microscopy. Cross section of the film sample [4].

It is known that surface roughness can make the interpretation of RBS spectra difficult [3]. In our study, to solve the problem of studying the influence of irregularities on the spectrum display, some samples were analyzed.

- functional environment for planar chemo-electronic converter,
- functional environment for volumetric chemo-electronic converter – tablet polished/ unpolished.

For comparison, the spectra were also considered:

- Polycrystalline ZrO₂ with polished surface;
- Single crystal ZrO₂ with juvenile surface.

A functional layer for producing planar chemo-electronic converters in the form of rounded drops containing monodisperse nanosized (7.5 μ m) particles of a solid solution of the ZrO₂ system -3 mol% Y₂O₃ (YSZ) in the PVA polymer matrix [4].

For the production of nanopowders used in chemical technology co-deposition with the use of physical effects [5].

Results

RBS is one of the most commonly used tools for depth pro- filing in various fields of physics, therefore the development of such a method is essential.

Figure 4a shows the results of RBS analysis of samples - functional environment for planar chemo-electronic converter and Figure 4b volumetric chemo-electronic converter – tablet polished/ unpolished.

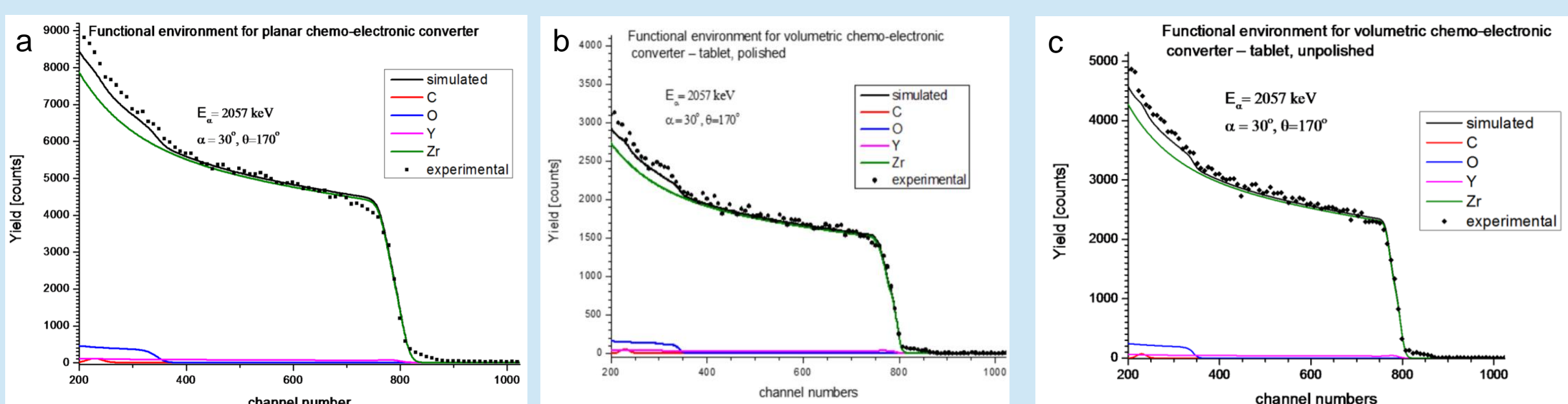


Figure 4. a - RBS spectrum of functional environment for planar chemo-electronic converter; b - RBS spectrum of volumetric chemo-electronic converter – tablet polished; c - RBS spectrum of volumetric chemo-electronic converter – tablet unpolished.

Using the RBS technique, **depth profiles** of samples with **different surface types** were studied, and it was concluded that this method is **well suited for studying samples consisting of nanosized particles**.

Further study of the coating under research should have significant support for science and industry.

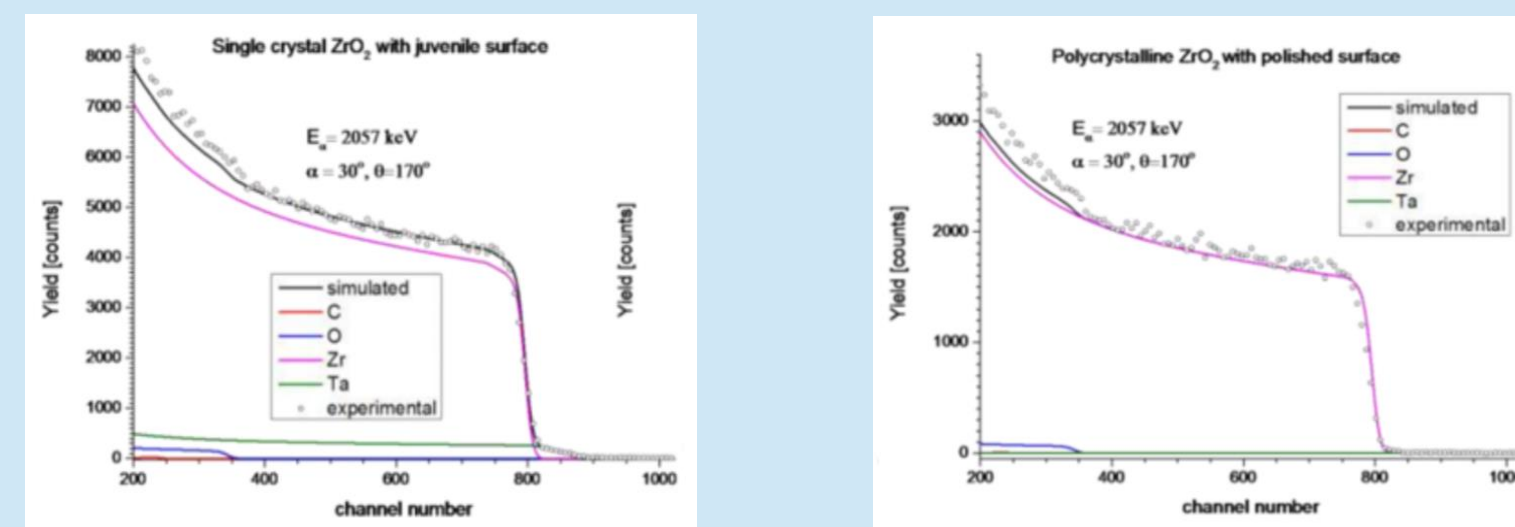


Figure 5. RBS spectrum of samples of the comparison.

The study was performed in the scope of the Project H2020/MSCA/RISE/SSHARE number 871284 project and RO-JINR Projects №.267/2020 item 25 and № 268/2020 item 51 and №.268/2020 item 57; Poland-JINR Project №75/2020 item 31.

Contact

Alisa A. Tatarinova
Joint Institute for Nuclear Research, FLNP,
Dubna, Russia
Chemistry, Materials and new technologies Department
Thin film technology laboratory, State University Dubna
Moscow region, Russia
Email: w99_9@yahoo.com
Phone: +7(950)-130-79-18

References

1. Techniques SMART Chart – Spectroscopy – RBS Available at: <https://www.eag.com/techniques/spectroscopy/rutherford-backscattering-spectrometry-rbs/> (accessed 13. 01. 20).
2. Schematic diagram of RBS facility Available at: <http://www.iuac.res.in/accel/paras/index.html> (accessed 13. 01. 20).
3. W. Chu, J. W. Mayer, and M. A. Nicolet, Backscattering Spectrometry (Academic, New York, 1978), p. 218.
4. Chemical-Electric Energy Conversion Effect in Zirconia Nanopowder Systems A. S. Doroshkevich, A. I. Lyubchik, A. V. Shilo, T. Yu. Zelenyak, V. A. Glazunovae, V. V. Burhovetskiy, A. V. Saprykina, Kh. T. Holmurodov, I. K. Nosolev, V. S. Doroshkevich, G. K. Volkova, T. E. Konstantinova, V. I. Bodnarchuk, P. P. Gladyshev, V. A. Turchenko, S. A. Sinyakina. (2017). Journal of Surface Investigation: X-ray, Synchrotron and Neutron Techniques Vol. 11, No. 3. - Pp. 523–529. DOI: 10.1134/S1027451017030053.
5. Konstantinova T.E., Danilenko I.A., Glazunova V.A., Volkova G.K., Gorban O.A. // Journal of nanoparticle research. 2011. V. 13. № 9. P. 4015 - 4023.