

INTRODUCTION

Magnesium (Mg) and Mg alloys considered as perspective materials for biodegradable implant development but fast and uncontrolled corrosion limited its clinical application. Plasma electrolytic oxidation (PEO) is a high-voltage electrochemical process that can provide protective coating over Mg implants. Alkaline silicate, sodium fluoride, and phosphate electrolytes are recently used for PEO due to impact on mechanical and protective properties of the coating. Additionally, PEO can provide porous surface topography that provide adequate environment for cell adhesion and proliferation.

AIM

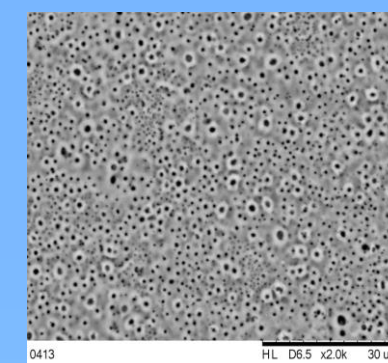
The objective of current study was to evaluate surface parameters of Mg after PEO in different alkali solutions.

METHODS

- Plasma electrolytic oxidation (PEO);
- Scanning electronic microscopy (SEM) (Surface morphologies);
- Contact angle (the energy of the surface, dispersion, hydrophilicity, hydrophobicity);
- Roughness.

SAMPLES
Mg alloy, 1 cm³

PEO modification treatment



Material characterization

SEM

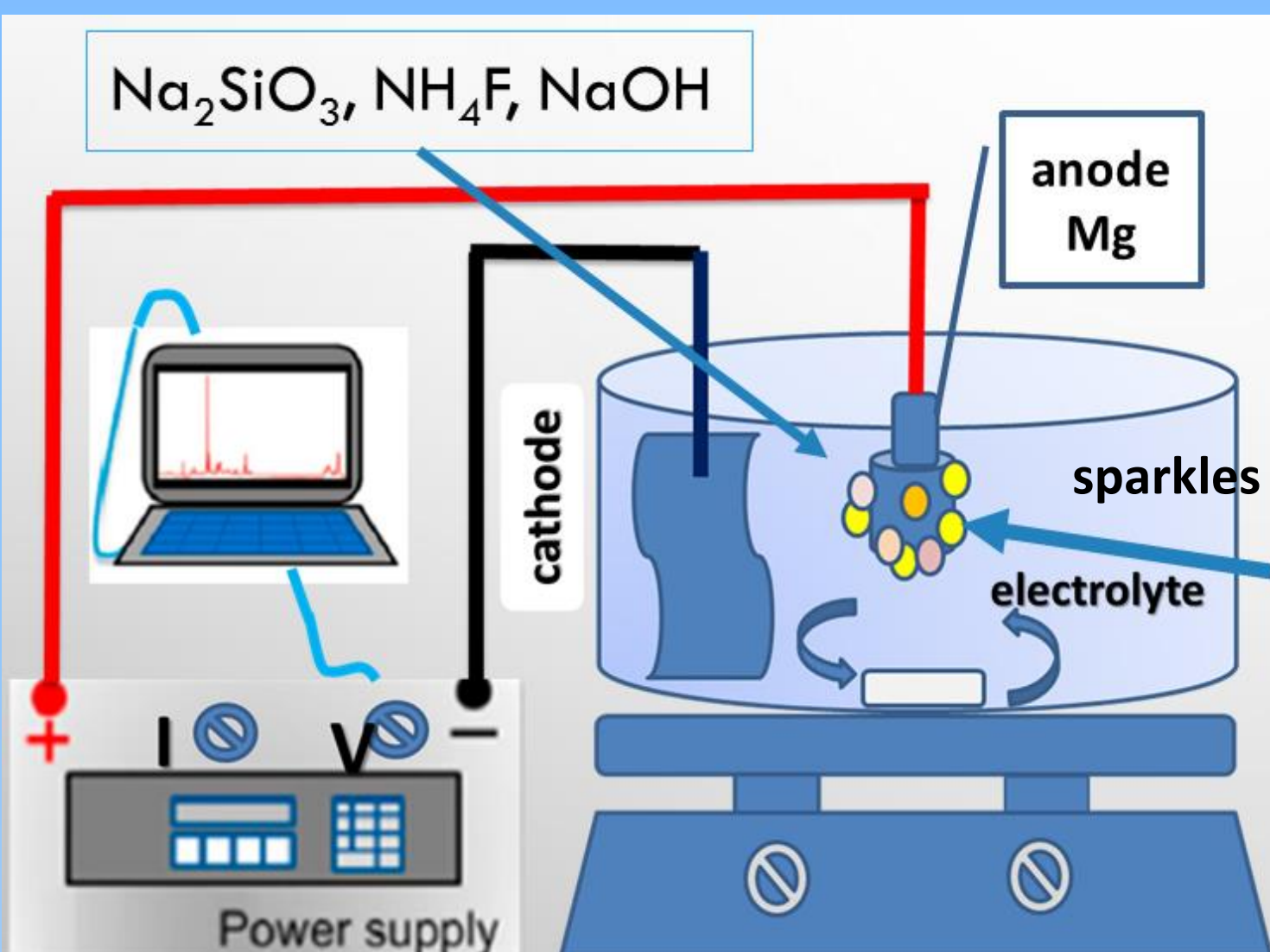
Contact angle (CA)

Roughness

PEO

Roughness

Contact angle



current density - 0.1 A cm⁻²
and final voltage - 200, 250, 275 and 300 V for 10 min



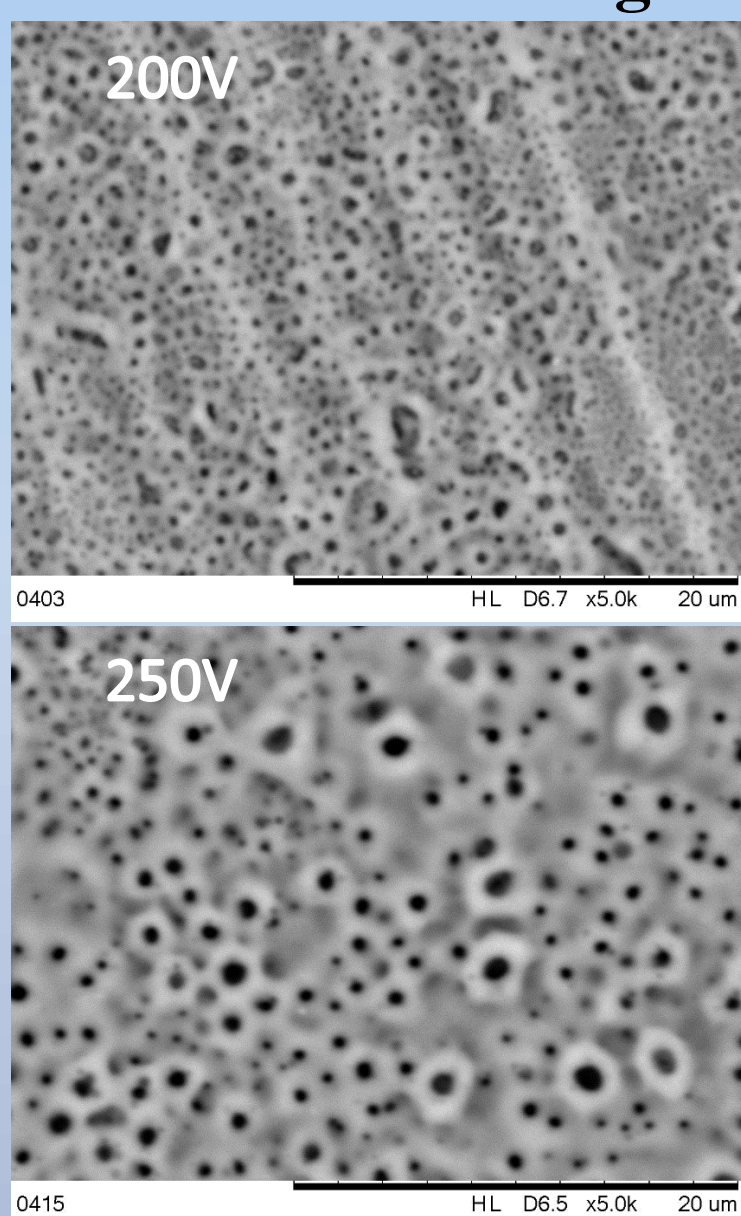
tactile stylus method using a surface roughness tester (Surftest SJ-301, Mitutoyo, Kawasaki, Kanagawa, Japan)



video-based optical contact angle measuring instrument (OCA 15 EC, Series GM-10-473 V-5.0, Data Physics, Filderstadt, Germany)

RESULTS

SEM images of the surface after PEO coatings

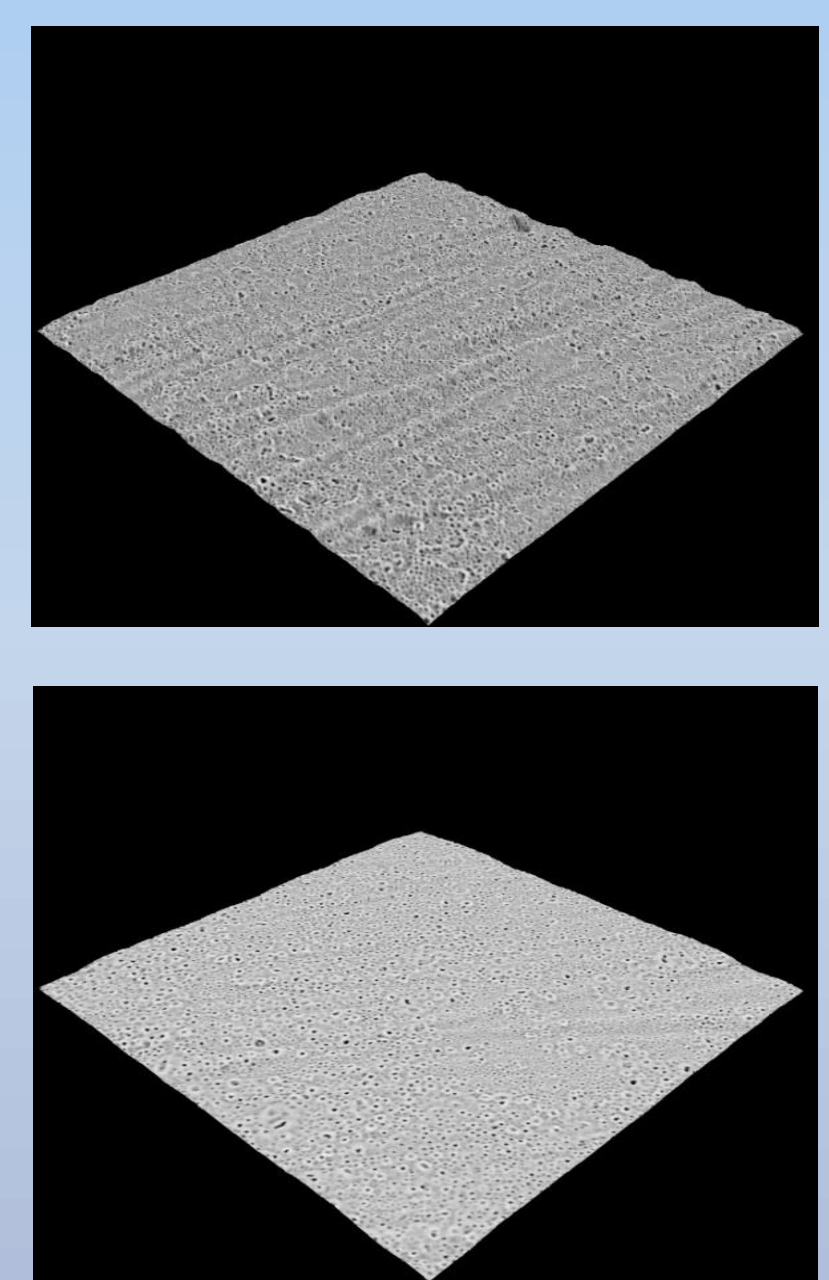
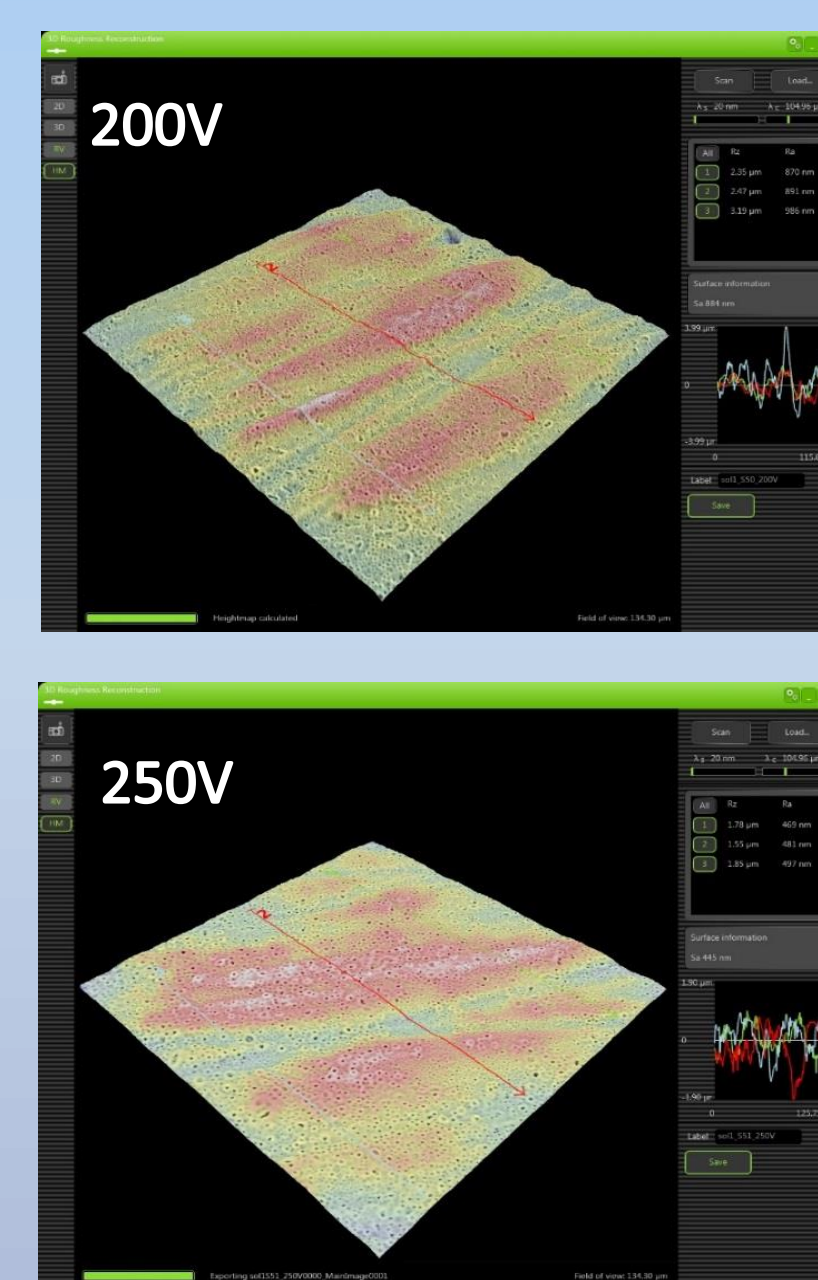


Characteristics of roughness and wettability of the coatings

Samples		Ra, μm	CA, °
Sol 1 Na ₂ SiO ₃ , NH ₄ F and NaOH	200V	0,43±0,02	26.22
	250V	0,39±0,01	22.7
Sol 2 Na ₂ SiO ₃ , NH ₄ F, and Ca(OH) ₂	250V	1,67±0,08	16.61
	300V	2,10±0,15 *	0
Sol 3 Na ₂ HPO ₄ and NaOH	250V	1,09±0,63	70.73
	275V	0,87±0,08	59.72
	300V	2,88±0,3 *	94.21

* - p<0.0001

The surface roughness of the surface after PEO coatings



The results showed statistically significant increases (p < 0.0001) the surface roughness in the case of Sol 2 300 V (2.10±0.15) and Sol 3 300 V (2.88±0.3) samples among obtained coatings.

The contact angle for the samples Sol 1 and Sol 2 characterized as the surface with high wettability. Samples Sol 3 refers to the surface with low wettability.

CONCLUSION

The Mg treated under Sol 2 with final voltage 300 V showed the higher roughness value and complete wettability. The hydrophilicity of obtained coating and high roughness made them suitable for biomedical applications.

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